Name \_\_\_\_\_\_ Put

# General Chemistry - Unit 8 Review

- 1. Vocabulary
  - a. mole
  - b. molar mass
  - c. Avogadro's number
  - d. empirical formula
  - e. molecular formula
- 2. Find the molar mass of the following:
  - a. KNO<sub>s</sub> \_\_\_\_\_

d. oxygen gas \_\_\_\_\_

b. (NH<sub>4</sub>)<sub>o</sub>CO<sub>5</sub> \_\_\_\_\_\_

e. calcium phosphate \_\_\_\_\_

c. Ag<sub>2</sub>CrO<sub>4</sub> \_\_\_\_\_\_

- f. lead (II) nitrate \_\_\_\_\_
- 3. Consider the masses of various hardware below.

Type	Mass (g)	Relative mass
Washer	1.74	
Hex nut	3.16	
Anchor	12 (2) (1)	3.00
Bolt	7.64	

- a. Do the calculations necessary to complete the table.
- Explain the connection between these calculations and the atomic masses in the Periodic Table.

# **V2 0 Chemistry Unit 8 Review**

F. Albert Cotton, Geoffrey
Wilkinson, Carlos A. Murillo, Manfred
Bochmann

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