The Mole Workbook Questions

W9.1	What is the mass of one mole of:
	(a) lithium atoms
	(b) sodium atoms
	(c) calcium atoms
	(d) iron atoms
	(e) silver atoms
	(f) lead atoms?
W9.2	What is the mass of:
	(a) 10 moles of chlorine atoms
	(b) 0.125 mole of bromine molecules
	(c) 0.25 mole of H ₂ O
	(d) 0.5 mole of sulfur molecules, S _x
	(e) 0.5 mole of HNO,
	(f) 0.5 mole of SO ₂ ions
	(g) 0.05 mole of oxygen molecules?
NAMES IN	Calculate the mass of:
	(a) 4 moles NaOH
	- 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
	(b) 0.5 mole CaCl _j
	(c) 2 moles Cl ₂
	(d) 0.05 mole MgO
	(e) 3 moles Fe ₂ O ₃
	(f) 0.1 mole C ₂ H ₈
	(g) 0.6 mole CaCO,
	(h) 0.3 mole NaCl

Skill Practice 31 Moles And Formulas Practice

Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas

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