Skill Practice 19



1. Write the symbol and charges for the following ions:

Example: calcium v Ca

- A/I pfluospfleide m
- B) magnesises w _____ C) rehidises w

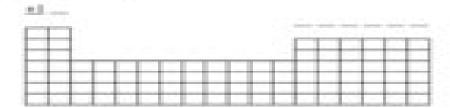
- D) fluoride = E) aluminum = F) sulfide =
- 2. Each of the following formulas is written incorrectly. Please rewrite them correctly. Hyamphr: Ca₂Cl = CaCl₂

 - A) BaS = ______ B) RbN = _____ C) LbC = _____
 - D) Al₂N₁ = _____ F) Mg/Br₂ = ____ F) O₂Al₂ = ____

- 3. Write the formulas for the compound formed by the combination of the following pairs of atoms. Example: nitrogen and magnesium a Ma-N-

A) redison and suffer = _______ B) alaminum and obtoring =

- C) phosphorus and calcium = D) burium and oxygen =
- For each of the formulas you wrote in question 3 above, give the names for the compounds. Example: Mg-N_c w magnesium mbride
- C) D)
- 5. In the blanks above each column, write the charge of the ions formed by atoms in that column. Note: group IA is done for you.



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Topic has three main goals 1 provide a platform for instructors of organic chemistry to showcase evidence based methods and educational theories they have utilized in their classrooms 2 build new and strengthen existing connections between educational researchers and practitioners and 3 highlight how people have used chemical education based research in their teaching practice. There are places in the literature dedicated for chemical education research CER however there is not a clear avenue for those that have changed their teaching methods based on published CER and report their experiences. Creating this article collection will foster collaboration between chemical education researchers and teachers of organic chemistry. This opportunity allows these instructors to share evidence based practices experiences challenges and innovative approaches from CER literature and beyond This Research Topic bridges discipline based education research and the scholarship of teaching and learning which will help advance organic chemistry education and improve student outcomes

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