

ATI TEAS 7 PRACTICE QUESTIONS AND ANSWERS.

- 1) Chemical bonding –
 - a. Uses electrons that are closest to the nucleus of the atoms bonding
 - b. Always uses electrons from only of the atoms involved
 - c. Uses all the electrons in all atoms involved
 - d. **Uses the valence electrons of all the atoms involved**
- 2) Which of the following would have the most dramatic effect in changing the reaction rate?
 - a. Decreasing the temperature by two-fold
 - b. Increasing the pressure by two-fold
 - c. **Decreasing the activation energy by three-fold**
 - d. Decreasing the concentration of reactants by two-fold
- 3) Group one in the periodic table has ____ valence electrons and is ____ reactive than Group two.
 - a. Zero; more
 - b. Zero; less
 - c. **One; less**
 - d. One; more
- 4) Enzymes are created from amino acid chains. As such, what might prevent the action of an enzyme?
 - a. A pH close to 7
 - b. **A high temperature**
 - c. The lack of hydrogen ions in solution
 - d. A lack of ATP in the cell
- 5) If a scientists want to determine the rate at which an enzyme works, what could he or she measure?
 - a. The rate at which the enzyme is degraded
 - b. The rate at which the product disappears
 - c. **The rate at which the reactants disappear**
 - d. The rate at which the products appears
- 6) The rate of a chemical reaction depends on all of the following except

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