# Specific Heat Lab

#### Procedure

- Place 150 mL of sand in a 250 mL beaker. Place 150 mL of water in a second 250 mL beaker.
- 2. Measure the initial temperature of the sand and of the water.
- 3. Place both of the beakers on a hot plate. Turn the hot plate on to setting 5.
- 4. Every minute for ten minutes, measure the temperature of the water and of the sand. Be sure that the thermometer does not touch the sides or bottom of the beaker.
- 5. Use tongs to remove the beaker of sand and the water from the hot plate.

#### <u>Data</u>

Time (minutes)	Temp. of sand	Temp. of water
1		
2		
3		
4		
5		
6		
7		
8		
9		
10		

#### Questions

- Define "specific heat".
- 2. If a substance absorbs heat easily, does it have a high specific heat or a low specific heat?
- 3. Which absorbs heat more easily, water or sand?
- 4. Which has a greater specific heat, water or sand?
- 5. What is the specific heat of water in J/g°C?

# **Specific Heat Lab Answers**

United States. National Bureau of Standards. Cryogenic Engineering Laboratory, Boulder, Colo

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