



## Trends in Atomic Radii

1. Describe and explain the trend in atomic radii as you go down group 2.
2. Describe and explain the trend in atomic radii as you go across period 3.
3. Describe and explain the trend in atomic radii as you go across period 4 from Scandium to Zinc.

## Radii Comparisons

4. Compare the radii of the following pairs, including both a description and an explanation in your answer.
  - a. Na and  $\text{Na}^+$
  - b. Li and  $\text{O}^-$
  - c.  $\text{Al}^{3+}$  and  $\text{Na}^+$
  - d.  $\text{H}^+$  and  $\text{H}^-$
  - e.  $\text{S}^{2-}$  and  $\text{Cl}^-$
  - f. F and  $\text{Na}^+$



Answers

# Teaching Transparency Atomic And Ionic Radii

**Society of Glass Technology**



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