

SI Chemistry

Chemical Reactions Lab

Purpose: to become more familiar with different types of chemical reactions by conducting several reactions, noting the results, and discussing observations with colleagues.

Procedure: At each of the nine stations, there are directions for you to follow-**including directions for waste disposal – FOLLOW ALL DIRECTIONS CAREFULLY!**

You should visit only six of the nine stations. You **MUST** visit stations 6, 8 and 9. You will then choose one station from stations 1 and 7, one station from stations 2 and 5 and one station from stations 3 and 4. You do not need to move through the stations in any particular order. You **MUST** wear your safety goggles at **ALL TIMES!**

At **each station**, answer the following questions in your lab notebook or on your tablet:

- a. How did you know a chemical reaction took place?
- b. Formula of reactant(s)
- c. Formula of product(s)
- d. Type of chemical reaction
- e. Write the balanced equation for the reaction (including all physical states).

Station 1: Get a strip of magnesium ribbon (about 2.5 cm long). Holding it with forceps, ignite the ribbon in a Bunsen burner flame. **Do not look directly at the burning magnesium!** **Looking directly at the burning magnesium could damage your eyes.** Hold the burning ribbon over a glass plate so that no burning magnesium lands on the lab bench.

Word equation: Magnesium reacts with oxygen to produce magnesium oxide.

Station 2: Place 10 drops of silver nitrate into a small test tube. Add 10 drops of sodium chloride to the same test tube.

Word equation: Aqueous silver nitrate reacts with aqueous sodium chloride to produce solid silver chloride and aqueous sodium nitrate.

Si Chemistry Chemical Reaction Lab Introduction

Minjie Lin



Si Chemistry Chemical Reaction Lab Introduction:

Air Force Research Resumés , **An Introductory Guide to EC Competition Law and Practice** Valentine Korah,1994 *Nuclear Science Abstracts* ,1974 Electrochemical Sensors Giuseppe Maruccio,Jagriti Narang,2022-02-01 *Electrochemical Sensors From Working Electrodes to Functionalization and Miniaturized Devices* provides an overview of the materials preparation and fabrication methods for biosensor applications The book introduces the field of electrochemistry and its fundamentals also providing a practical overview of working electrodes as key components for the implementation of sensors and assays Features covered include the prompt transfer of electrons favorable redox behavior biocompatibility and inertness in terms of electrode fouling Special attention is dedicated to analyzing the various working materials systems for electrodes used in electrochemical cells such as gold carbon copper platinum and metal oxides This book is suitable for academics and practitioners working in the disciplines of materials science and engineering analytical chemistry and biomedical engineering Introduces key concepts for electrochemistry and biosensors Reviews the most common and emerging materials based electrodes for sensor applications including gold carbon platinum and metal oxides Discusses both macro and miniaturized electrodes including their cleaning engineering fabrication examples of working biosensors and advantages and disadvantages Scientific and Technical Aerospace Reports ,1995-05 **Energy Research Abstracts** ,1993 Solar Energy Update ,1981 **ERDA Energy Research Abstracts** United States. Energy Research and Development Administration,1977 *Subject Index to Unclassified ASTIA Documents* Defense Documentation Center (U.S.),1960 Air Force Research Resumés ,1963 **Calendar** University of Saskatchewan,1929 **Catalog of Technical Reports** United States. Dept. of Commerce. Office of Technical Services, **NASA Tech Briefs** ,1988 Index to ... NASA Tech Briefs ,1989 *ERDA Energy Research Abstracts* United States. Energy Research and Development Administration. Technical Information Center,1977 *Undergraduate Catalog* ,1961 Includes a directory of undergraduate programs academic calendars course descriptions listing of faculty campus maps and general information on topics such as academic support services academic resources admissions registration financial aid student services and student activities Also provides a brief overview including graduation requirements policies and special learning opportunities of the following University colleges College of Agricultural Food and Environmental Sciences College of Architecture and Landscape Architecture College of Biological Sciences Program in Dental Hygiene College of Education and Human Development General College College of Human Ecology College of Liberal Arts Curtis L Carlson School of Management Program in Medical Technology Program of Mortuary Science College of Natural Resources School of Nursing ROTC Institute of Technology and University College *U.S. Government Research Reports* ,1964 **ERDA Research Abstracts** United States. Energy Research and Development Administration,1976 **Learning Directory** ,1972 *Technical Abstract Bulletin* ,1964

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Table of Contents Si Chemistry Chemical Reaction Lab Introduction

1. Understanding the eBook Si Chemistry Chemical Reaction Lab Introduction
 - The Rise of Digital Reading Si Chemistry Chemical Reaction Lab Introduction
 - Advantages of eBooks Over Traditional Books
2. Identifying Si Chemistry Chemical Reaction Lab Introduction
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Si Chemistry Chemical Reaction Lab Introduction
 - User-Friendly Interface
4. Exploring eBook Recommendations from Si Chemistry Chemical Reaction Lab Introduction

- Personalized Recommendations
- Si Chemistry Chemical Reaction Lab Introduction User Reviews and Ratings
- Si Chemistry Chemical Reaction Lab Introduction and Bestseller Lists
- 5. Accessing Si Chemistry Chemical Reaction Lab Introduction Free and Paid eBooks
 - Si Chemistry Chemical Reaction Lab Introduction Public Domain eBooks
 - Si Chemistry Chemical Reaction Lab Introduction eBook Subscription Services
 - Si Chemistry Chemical Reaction Lab Introduction Budget-Friendly Options
- 6. Navigating Si Chemistry Chemical Reaction Lab Introduction eBook Formats
 - ePub, PDF, MOBI, and More
 - Si Chemistry Chemical Reaction Lab Introduction Compatibility with Devices
 - Si Chemistry Chemical Reaction Lab Introduction Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Si Chemistry Chemical Reaction Lab Introduction
 - Highlighting and Note-Taking Si Chemistry Chemical Reaction Lab Introduction
 - Interactive Elements Si Chemistry Chemical Reaction Lab Introduction
- 8. Staying Engaged with Si Chemistry Chemical Reaction Lab Introduction
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Si Chemistry Chemical Reaction Lab Introduction
- 9. Balancing eBooks and Physical Books Si Chemistry Chemical Reaction Lab Introduction
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Si Chemistry Chemical Reaction Lab Introduction
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Si Chemistry Chemical Reaction Lab Introduction
 - Setting Reading Goals Si Chemistry Chemical Reaction Lab Introduction
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Si Chemistry Chemical Reaction Lab Introduction

- Fact-Checking eBook Content of Si Chemistry Chemical Reaction Lab Introduction
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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