1.10 Assume the following: 
$$\rho := 13.5 \frac{gm}{cm^3}$$
  $g := 9.8 \frac{m}{s^2}$ 

$$P := 400bar \qquad h := \frac{P}{cm^3} \qquad h = 302.3 m \quad Ans.$$

1.11 The force on a spring is described by:  $F = K_s x$  where  $K_s$  is the spring constant. First calculate K based on the earth measurement then  $g_{Mars}$  based on spring measurement on Mars. On Earth:

$$g_{Mars} := 0.40em$$
  $F_{Mars} := K \cdot x$   $F_{Mars} = 4 \times 10^{-3} \text{ mK}$   $g_{Mars} := \frac{F_{Mars}}{mass}$   $g_{Mars} = 0.01 \frac{mK}{kg}$  Ans.

1.12 Given: 
$$\frac{d}{dz}P = -\rho \cdot g$$
 and:  $\rho = \frac{M \cdot P}{R \cdot T}$  Substituting:  $\frac{d}{dz}P = -\frac{M \cdot P}{R \cdot T} \cdot g$ 

Separating variables and integrating: 
$$\int_{P_{sea}}^{P_{Denver}} \frac{1}{P} dP = \int_{0}^{z_{Denver}} -\left(\frac{M \cdot g}{R \cdot T}\right) dz$$
After integrating: 
$$\ln\left(\frac{P_{Denver}}{P_{sea}}\right) = \frac{-M \cdot g}{R \cdot T} \cdot z_{Denver}$$

Taking the exponential of both sides and rearranging: 
$$P_{Denver} = P_{sea} \cdot e^{\left(\frac{-M \cdot g}{R \cdot T} \cdot z_{Denver}\right)}$$
 
$$P_{sea} := 1 atm \qquad M := 29 \frac{gm}{mol} \qquad g := 9.8 \frac{m}{s^2}$$

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Water Quality Criteria, 1972 National Research Council (U.S.). Committee on Water Quality Criteria, 1972

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