

1. The table shows some information about solids, liquids and gases.

Complete the table.

state	distance between particles	movement of particles	forces between particles	shape
solid	close together	fixed shape
liquid	close together	move slowly in all directions	shape of container
gas	move quickly in all directions	very weak

2. This question is about liquids.

(a) When a liquid is put into a beaker it takes on the shape of the beaker.

Why do liquids take on the shape of the beaker?

Tick (✓) the boxes next to **two** correct answers.

Liquids have no fixed shape.

☐

Liquids have no fixed volume.

☐

Liquid particles only vibrate.

☐

The particles in a liquid expand.

☐

The particles in a liquid move around.

☐

(b) Pierre investigates the boiling point of water.

- He heats up some pure water in a beaker until it boils.
- Pierre measures the boiling point of the pure water.
- He does this two more times using new samples of pure water.
- Pierre then does this three more times using salt water.

The table shows some of his results.

type of water	boiling point in °C		
pure water	100	100	100
salt water	102	103	

(i) There is one result missing. The thermometer shows Pierre's missing result.

States Of Matter Section 13 1 Gases Worksheet Answers

Peipei Pang



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Table of Contents States Of Matter Section 13 1 Gases Worksheet Answers

1. Understanding the eBook States Of Matter Section 13 1 Gases Worksheet Answers
 - The Rise of Digital Reading States Of Matter Section 13 1 Gases Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying States Of Matter Section 13 1 Gases Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an States Of Matter Section 13 1 Gases Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from States Of Matter Section 13 1 Gases Worksheet Answers
 - Personalized Recommendations
 - States Of Matter Section 13 1 Gases Worksheet Answers User Reviews and Ratings
 - States Of Matter Section 13 1 Gases Worksheet Answers and Bestseller Lists
5. Accessing States Of Matter Section 13 1 Gases Worksheet Answers Free and Paid eBooks
 - States Of Matter Section 13 1 Gases Worksheet Answers Public Domain eBooks
 - States Of Matter Section 13 1 Gases Worksheet Answers eBook Subscription Services
 - States Of Matter Section 13 1 Gases Worksheet Answers Budget-Friendly Options

6. Navigating States Of Matter Section 13 1 Gases Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - States Of Matter Section 13 1 Gases Worksheet Answers Compatibility with Devices
 - States Of Matter Section 13 1 Gases Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of States Of Matter Section 13 1 Gases Worksheet Answers
 - Highlighting and Note-Taking States Of Matter Section 13 1 Gases Worksheet Answers
 - Interactive Elements States Of Matter Section 13 1 Gases Worksheet Answers
8. Staying Engaged with States Of Matter Section 13 1 Gases Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers States Of Matter Section 13 1 Gases Worksheet Answers
9. Balancing eBooks and Physical Books States Of Matter Section 13 1 Gases Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection States Of Matter Section 13 1 Gases Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine States Of Matter Section 13 1 Gases Worksheet Answers
 - Setting Reading Goals States Of Matter Section 13 1 Gases Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of States Of Matter Section 13 1 Gases Worksheet Answers
 - Fact-Checking eBook Content of States Of Matter Section 13 1 Gases Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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