

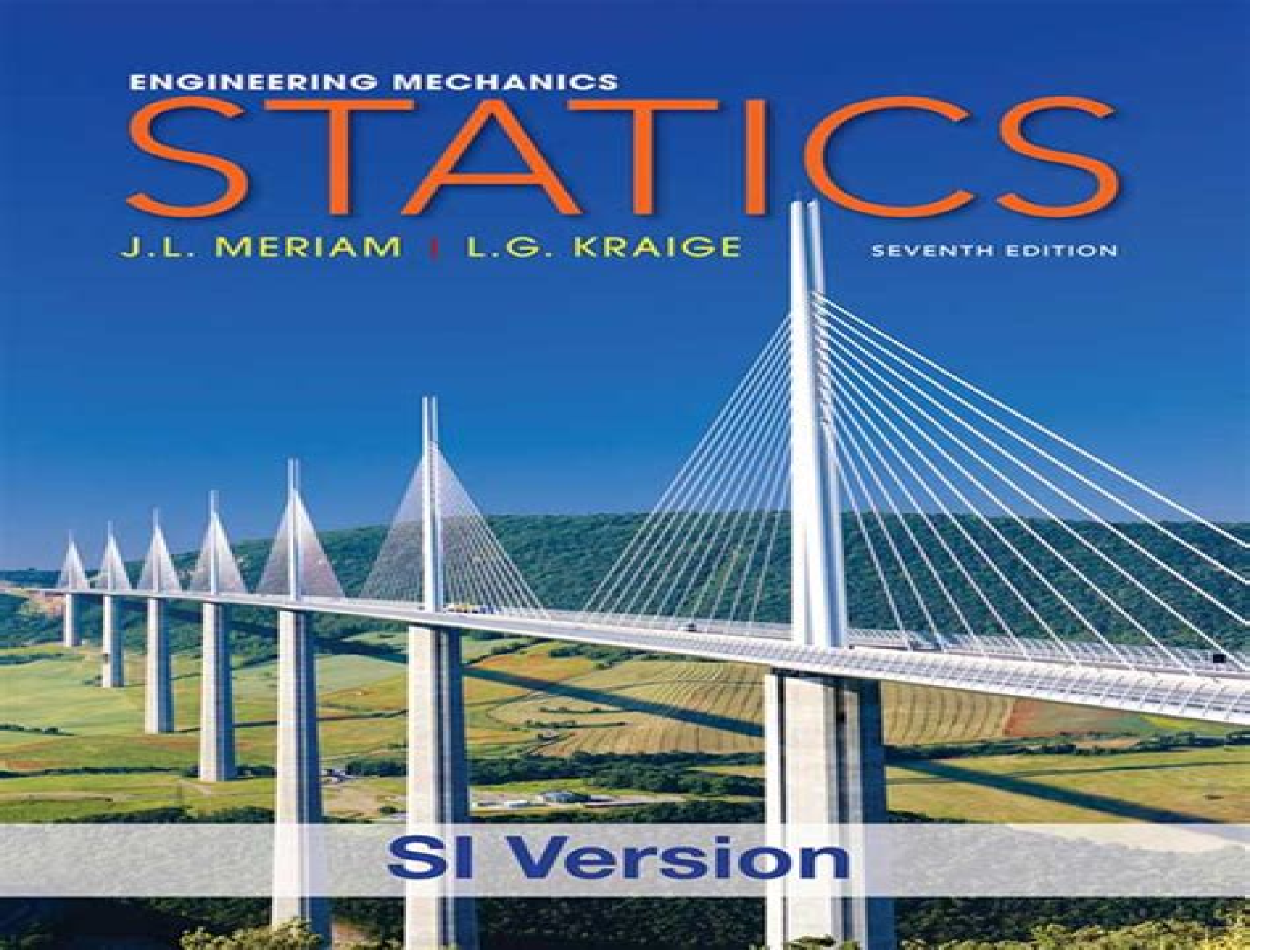
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 oh $\text{HCO}_2\text{H} + 2\text{H}^+ + 4\text{e}^- \rightarrow \text{H}_2\text{C}_2\text{O}_4$ oxidation $4\text{Fe} + 8\text{H}^+ + \text{MnO}_4^- \rightarrow 4\text{Fe}^{2+} + 4\text{H}_2\text{O} + \text{Mn}^{2+}$ reduction $12\text{H}^+ + 5\text{CH}_3\text{OH} + 4\text{MnO}_4^- \rightarrow 5\text{HCO}_2\text{H} + 4\text{Mn}^{2+} + 11\text{H}_2\text{O}$ d in acid
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