



## PARTICLE THEORY AND STATES OF MATTER

What is the basic structure of matter? Understanding the nature of matter and the arrangement of particles is fundamental. The arrangement and movement of particles explain various properties of the matter.

**Particle Theory** is the theory that all matter is made up of tiny particles. These particles have three characteristics of particles: they are very small, they are in constant motion, and they have spaces between them. The arrangement and movement of particles explain various properties of the matter.

**States of Matter** - Matter can exist in three primary states: solid, liquid, and gas. Each state has different properties. Solids have a definite shape and volume. Liquids have a definite volume but no definite shape. Gases have neither a definite shape nor a definite volume. The particles in solids are closely packed, in liquids they are less closely packed, and in gases they are far apart.

**Changes of State** - Matter can change from one state to another. These changes are called changes of state. They are caused by the addition or removal of heat. For example, melting is the change from solid to liquid, and boiling is the change from liquid to gas. These changes are reversible.



1. How does the particle theory explain the expansion of solids when heated?

10. Complete the diagram showing the arrangement of particles in each state of matter. Label the particles and the spaces between them.

9. What is the change of state when a solid melts?

8. Why is the density of matter different in different states?

7. How does the particle theory explain the contraction of gases when cooled?

1. What is particle theory?

2. List all the states of matter. Describe each state briefly.

3. Give two examples of changes of state. Explain each change.

4. The particles in a solid are closely packed. Why do they not move freely?

5. What happens when a solid is heated?

6. How does the particle theory explain the expansion of gases when heated?

# States Of Matter Guided Reading Answer Key

**George Graybill**



## **States Of Matter Guided Reading Answer Key:**

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**Prentice Hall Chemistry**,2000 Matter and Energy ,2005 **Resources in Education** ,1992 **Research in Education** ,1968 It Does

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