

States Of Matter

Matter is all around us and can be found in one of three major states: solid, liquid or gas

Solids

- keep their shape and can be held
- do not flow or spread out
- keep the same volume
- can be cut or shaped



Liquids

- will take the shape of a container
- can be poured and will flow
- keep the same volume
- can not be held easily



Gases

- will spread out and fill a container
- volume can increase or decrease
- can flow through pipes
- are often invisible



Particles

Particles are tiny bits of matter that make up all solids, liquids and gases



Particles are very close together and can't move easily



Particles are close together but can move around



Particles are far apart and free to move around

Changing State



Evaporation

When a liquid slowly turns into a gas (without boiling), this is called evaporation. It is why puddles disappear and your clothes dry outside on the washing line



Liquids evaporate at a faster rate in warm temperatures or on windy days

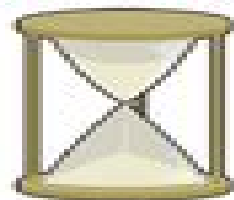
Condensation

Condensation is when a gas turns back into a liquid. It is why a cold drink gets wet on the outside and why a mirror steams up during a hot shower (from the water vapour in the air)



Smoke (from a fire) is a collection of tiny solid, liquid and gas particles

Solid or Liquid?

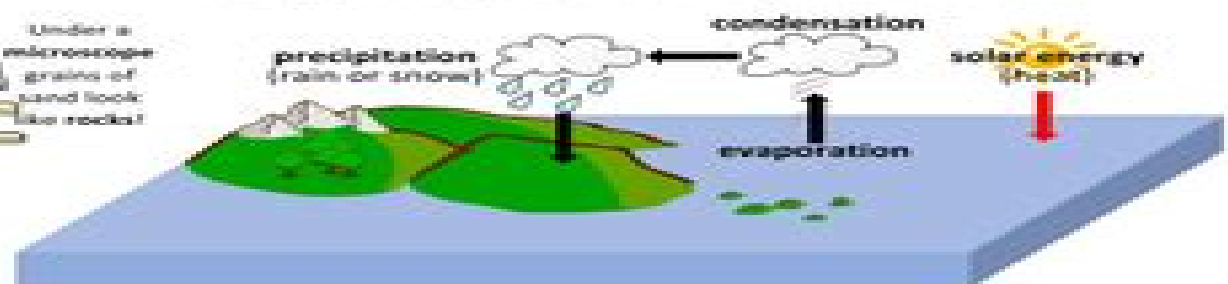


Under a microscope grains of sand look like rocks!

Sand, salt and flour behave a bit like a liquid because they can be poured and take the shape of a container. However, they are actually very small solids that keep their shape and volume

The Water Cycle

The water cycle is the journey of water as it evaporates and condenses



Rain isn't salty because only the water evaporates leaving the salt behind

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