



**ZIMBABWE SCHOOL EXAMINATIONS COUNCIL**  
General Certificate of Education Ordinary Level

**COMPUTER SCIENCE**  
**PAPER 1 Multiple Choice**

**4021/1**

**SPECIMEN PAPER**

**1 hour**

*Additional materials:*

*Multiple Choice answer sheet*

*Soft clean eraser*

*Soft pencil (type B or HB is recommended.)*

**TIME** 1 hour

**INSTRUCTIONS TO CANDIDATES**

Do **not** open this booklet until you are told to do so.

Write your name, centre number and candidate number on the answer sheets in the spaces provided unless this has already been done for you.

Read very carefully the instructions on the answer sheet.

**INFORMATION FOR CANDIDATES**

Each correct answer will score **one** mark. A mark will not be deducted for a wrong answer. Any rough working should be done in this booklet.

There are **forty** questions in this paper. Answer **all** questions. For each question there are four possible answers **A, B, C** and **D**. Choose the one you consider correct and record choice in soft pencil on the separate answer sheet provided.

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# Zimsec Examination Papers June 2015

**Robert R. Redfield**



**Zimsec Examination Papers June 2015:**

## Reviewing **Zimsec Examination Papers June 2015**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Zimsec Examination Papers June 2015**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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web solution a titrant volume 0 ml the solution pH is due to the acid ionization of HCl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the pH of the solution is then  $pH = -\log 0.100 = 1.000$  pH  $\log 0.100 = 1.000$  b titrant volume 12.50 ml

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web titration curves source openstax chemistry atoms first 2e strong acid strong base titrations when titrating a strong acid such as hydrochloric acid with sodium hydroxide you are reacting the HCl directly with NaOH  $HCl(aq) + NaOH(aq) \rightarrow H_2O(l) + NaCl(aq)$

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web ab acid base titration curves purpose the purpose of this lab is to determine the shape of acid base titration curves theoretically and experimentally and to use an acid base titration to measure the effectiveness of antacid tablets summary a titration curve is the plot of the  $pH$  value of a solution vs the amount of titrant added

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| Volume of NaOH added (mL) | $pH$ of the solution |
|---------------------------|----------------------|
| 4                         | 6.4                  |
| 6                         | 4.6                  |
| 4                         | 6.5                  |
| 9                         | 5.9                  |
| 5                         | 9.5                  |
| 9                         | 5.9                  |

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