

PHYSICAL CHEMISTRY

(Complete Syllabus)

SECTION - A

Straight Objective Type

This section contains 220 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer :

- The mass of oxygen required to produce enough CO which completely reduces x gm of Fe_2O_3 . Suppose molecular weight of $\text{O}_2 = y$, molecular weight of $\text{Fe}_2\text{O}_3 = z$
(1) $\frac{2xz}{3y}$ (2) $\frac{3xy}{2z}$
(3) $\frac{2y}{3xz}$ (4) $\frac{3z}{2xy}$
- The number of lone pair of electrons present in 2.56 gm of orthorhombic sulphur will be
(1) 0.02 N_0 (2) 0.16 N_0
(3) 0.32 N_0 (4) 0.08 N_0
- A 10.0 g of sample of mixture of calcium chloride and sodium chloride is treated with Na_2CO_3 to precipitate the calcium as calcium carbonate. This CaCO_3 is heated to convert all calcium to CaO and final mass of CaO is 1.62 gms. The % by mass of CaCl_2 in the original mixture is
(1) 15.2%
(2) 32.1%
(3) 21.8%
(4) 11.07%
- 1000 ml of oxygen at STP were passed through an ozoniser and the resulting volume was 888 ml at STP. This quantity of ozonised oxygen is passed through excess of KI solution. The weight of I_2 liberated is
(1) 25.4 g (2) 12.7 g
(3) 2.54 g (4) 1.27 g
- x_1 ml solution having normality 1 is diluted to get a solution of normality y_2 . What would be volume required for dilution?
(1) $x_1 \left(1 - \frac{y_1}{y_2}\right)$ (2) $x_1 \left(\frac{y_2}{y_1} - 1\right)$
(3) $x_1 \left(\frac{y_1}{y_2} - 1\right)$ (4) $x_1 \left(1 + \frac{y_1}{y_2}\right)$
- 1.625 g of an interhalogen compound ICl_x was subjected to electric discharge which split it into constituents and brought to STP. It occupied 112 ml. The interhalogen compound is
(1) ICl (2) ICl_3
(3) ICl_5 (4) ICl_7
- If 12.25 g of a sample of KClO_3 of 75% purity and rest vapourisable impurities is strongly heated to give KCl , KClO_4 and 0.32 gm O_2 . The mass of residue obtained is
(1) 1.769 g (2) 7.096 g
(3) 5.327 g (4) 8.85 g
- A 2 g sample of xenon reacts with fluorine. The mass of compound produced is 3.158 g. The empirical formula of the compound is
(1) XeF_2 (2) XeF_4
(3) XeF_6 (4) XeF_8
- 1 g of carbonate of metal was dissolved in 25 ml of 1 N HCl . The resulting liquid required 5 ml of 1 N NaOH for neutralisation. The equivalent weight

Zimsec Questions On Advanced Level Physical Chemistry

Peipei Pang



Zimsec Questions On Advanced Level Physical Chemistry:

Sociology A-Level (ZIMSEC) Past Exam Questions and Model Answers David Chitate ,2024-06-06 This book contains the most comprehensive question bank and model answers for ZIMSEC A Level exam questions It also includes syllabus review notes and exercises Examiners provide observations and tips and point out common errors that students make when answering questions If you use this book faithfully you can t fail and the Grade A is very much within your reach

School Science Practical Work in Africa Umesh Ramnarain,2020-06-01 School Science Practical Work in Africa presents the scope of research and practice of science practical work in African schools It brings together prominent science educators and researchers from Africa to share their experience and findings on pedagogical innovations and research informed practices on school science practical work The book highlights trends and patterns in the enactment and role of practical work across African countries Practical work is regarded as intrinsic to science teaching and learning and the form of practical work that is strongly advocated is inquiry based learning which signals a definite paradigm shift from the traditional teacher dominated to a learner centered approach The book provides empirical research on approaches to practical work contextual factors in the enactment of practical work and professional development in teaching practical work This book will be of great interest to academics researchers and post graduate students in the fields of science education and educational policy

Science Teachers Association of Nigeria Science Teachers Association of Nigeria. Annual Conference,2002* **Proceedings of the ... Annual Conference and ... Conference of CASTME Africa** Science Teachers Association of Nigeria. Conference,2002 **Physical Chemistry** M.M.L. Oblitas,1971 Physical Chemistry: A Level Multiple Choice Questions Margaret Marion Langley Oblitas,1970-01-01 **Revision Notes for Advanced Level Physical Chemistry** David Gwyn Davies,1968-01-01 **Multiple Choice Questions in A-level Chemistry** B. P. Bourne,1969 **Physical Chemistry** Robert S. Lowrie,1972 *Physical Chemistry* M.M.L. Oblitas,1977 Revision Notes in Advanced Level Chemistry: Physical chemistry Albert Holderness,1978 Using Your Chemistry Comprehension Questions for Advanced Level Peter Scott,Chiltern Society Staff,1994-12-01 **McGraw-Hill's 500 Physical Chemistry**

Questions: Ace Your College Exams Richard H. Langley,2012-12-12 500 Ways to Achieve Your Best Grades We want you to succeed on your physical chemistry midterm and final exams That s why we ve selected these 500 questions to help you study more effectively use your preparation time wisely and get your best grades These questions and answers are similar to the ones you ll find on a typical college exam so you will know what to expect on test day Each question includes explanations for right and wrong answers for your full understanding of the concepts Whether you have been studying all year or are doing a last minute review McGraw Hill s 500 Physical Chemistry Questions will help you achieve the final grade you desire Sharpen your subject knowledge and build your test taking confidence with 500 essential physical chemistry questions with answers Explanations for every answer Coverage of physical chemistry from ethical theory to epistemology **Physical**

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Revision notes in advanced level chemistry. Vol. 3 Albert Holderness,1968 *Problems and Solutions:* Kalidas C,2020-12

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Salient Features This book covers all the main topics of physical chemistry at the undergraduate and post graduate levels includes essential theoretical aspects required for problem solving provides 544 worked out problems 351 practice problems with solutions 508 multiple choice questions with answers covers the CBCS UGC syllabus authored by eminent professors from IIT aids preparation for competitive examinations such as CSIR UGC NET GATE and JAM includes solved question papers of CSIR UGC NET GATE and JAM provides access to an App for revision on your Android mobile phone

Objective Questions in Advanced Level Chemistry Edgar William Jenkins,1979 *Physical Chemistry* C. G. Silcocks,1966

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