

6. Write the equilibrium expression in terms of partial pressures (K_p) for each of the following reactions.

Rate the reactions in order of their increasing tendency to proceed toward completion:

b c d a



$K_p = 1 \times 10^{28} \text{ atm}$

$$K_p = \frac{(P_{\text{N}_2})^2 (P_{\text{H}_2\text{O}})^6}{(P_{\text{NH}_3})^4 (P_{\text{O}_2})^3}$$



$K_p = 5 \times 10^{-11}$

$$K_p = \frac{(P_{\text{NO}})^2}{(P_{\text{N}_2})(P_{\text{O}_2})}$$



$K_p = 1 \times 10^{-13}$

$$K_p = \frac{(P_{\text{H}_2})(P_{\text{F}_2})}{(P_{\text{HF}})^2}$$



$K_p = 4.7 \times 10^{-4} \text{ atm}$

$$K_p = \frac{(P_{\text{NO}})^2 (P_{\text{Cl}_2})}{(P_{\text{NOCl}})^2}$$

A Question That You Should Be Able To Answer:
Why don't the K_p 's in (b) and (c) have units?

(b) $\frac{\text{atm}^2}{\text{atm}^2} = 1$ (c) $\frac{\text{atm}^2}{\text{atm}^2} = 1$

7. (a) Write the K_c expression for
 $2\text{SO}_2(\text{g}) + \text{O}_2(\text{g}) \rightleftharpoons 2\text{SO}_3(\text{g})$

$$K_c = \frac{[\text{SO}_3]^2}{[\text{SO}_2]^2 [\text{O}_2]}$$

Calculate the value of K_c :

At equilibrium: $[\text{SO}_2] = 1.50 \text{ M}$
 $[\text{O}_2] = 1.25 \text{ M}$
 $[\text{SO}_3] = 3.50 \text{ M}$

$$= \frac{(3.50)^2}{(1.50)^2 (1.25)} = 4.355$$

$$= 4.36 \text{ M}^{-1}$$

- (b) If we reverse the equation, it is:



Write the K_c expression for this equation and calculate the new value of K_c :

$$K_c' = \frac{[\text{SO}_2]^2 [\text{O}_2]}{[\text{SO}_3]^2} = \frac{(1.5)^2 (1.25)}{(3.50)^2} = .22959$$

How does the expression and the value of K_c in 7(b) compare with those in 7(a)?

$$K_c' = \frac{1}{K_c}$$

- (c) If we now multiply all of the coefficients by $\frac{1}{2}$:



Write the K_c expression for this equation and calculate the new value of K_c :

$$K_c'' = \frac{[\text{SO}_2] [\text{O}_2]^{1/2}}{[\text{SO}_3]} = \frac{(1.5) (1.25)^{1/2}}{(3.50)} = .479$$

How do they compare with 7(b)?

$$K_c'' = \sqrt{K_c'}$$

- (d) What would happen to the K_c expression and its value if we doubled the coefficients?

$$K_c''' = (K_c')^2$$

Summarize:

Equation	K_c expression & Value
doubled	SQUARE
reversed	inverse
halved	SQ. ROOT

8. Consider an equilibrium that occurs in two steps:



- (a) Write the overall reaction.

(b) How do the K_c 's for the two steps (K_{c1} & K_{c2}) relate to the K_c of the overall reaction (K_c)?

$$K_{c1} \cdot K_{c2} = K_c$$

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RC Schank



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Millikan's School: A History of the California Institute of Technology

Judith Goodstein, 2020-10-19

In November 1891 wealthy former abolitionist and Chicago politician Amos Throop founded a thoroughly undistinguished small college in Pasadena California which he named after himself Millikan's School is the history of this institution that stands today at the pinnacle of world academics with 300 full time faculty nearly 1 000 undergraduate 1 250 graduate students and 39 Caltech and alumni Nobel Prize recipients Although Amos Throop the name of the college was changed to Caltech in 1920 could not have realized the importance of geography the fact that Pasadena lay at the foot of Mount Wilson was central to its success astronomer George Ellery Hale built his telescope there in 1902 the finest at that time in the world Later Hale joined the board of trustees of the struggling school and persuaded Arthur Amos Noyes former president of MIT and the nation's leading physical chemist to join him in Pasadena The third member of Caltech's founding troika was renowned physicist Robert A Millikan from the University of Chicago The dedication of Caltech in 1920 and the proclamation of what it stood for in science and education set the stage for Millikan who functioned as the school's president to bring the best and the brightest from all over the world Theodore von Kármán in aeronautics Thomas Hunt Morgan in biology Paul Sophus Epstein in physics Beno Gutenberg in seismology Linus Pauling in chemistry to Pasadena to work in an ever larger number of areas in science and technology The book also covers the funding planning and construction of the 200 inch telescope on Palomar Mountain Willy Fowler's work in nuclear astrophysics and the wartime rocket experiments that grew into the Jet Propulsion Laboratory JPL today the world leader in deep space exploration Millikan's School presents an interesting and thoroughly reliable account of the astonishing change over a period of a few years of a small technical school in Pasadena California into one of the world's leading scientific institutions Linus Pauling

In Millikan's School Judith Goodstein tells the remarkable story of the rise of Caltech She details how Millikan aided by Hale and Arthur Amos Noyes America's leading physical chemist and another of Hale's inspired acquisitions took a former trade school and forged from it a grandiose university among the orange groves It would be impossible while reading Goodstein's lively account not to be impressed by the energy drive and boundless enthusiasm of men like Millikan Hale and Noyes who had the bare faced audacity to set about building an institute to rival the cream of the universities of Europe and America Marcus Chown New Scientist Goodstein's story is first and foremost the tale of three men the astronomer George Ellery Hale the chemist Alfred Noyes and the physicist Robert Millikan It is the story of their attempts to transform an undistinguished little school founded in 1891 into a world class scientific establishment A useful book Tony Rothman Science In Millikan's School the story of Throop University's transformation into Caltech is told with precision Judith Goodstein's history offers a quick tour of the landmarks of science in the mid 20th Century and a glance at how pure science puts itself at the service of government commerce and the military Goodstein approaches her subject with a healthy sense of humor and an acute sense of academic

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what life under Prohibition was like and how such an unprecedented degree of government interference in the private lives of Americans changed the country forever Writing with both wit and historical acuity Okrent reveals how Prohibition marked a confluence of diverse forces the growing political power of the women s suffrage movement which allied itself with the antiliquor campaign the fear of small town native stock Protestants that they were losing control of their country to the immigrants of the large cities the anti German sentiment stoked by World War I and a variety of other unlikely factors ranging from the rise of the automobile to the advent of the income tax Through it all Americans kept drinking going to remarkably creative lengths to smuggle sell conceal and convivially and sometimes fatally imbibe their favorite intoxicants Last Call is peopled with vivid characters of an astonishing variety Susan B Anthony and Billy Sunday William Jennings Bryan and bootlegger Sam Bronfman Pierre S du Pont and H L Mencken Meyer Lansky and the incredible if long forgotten federal official Mabel Walker Willebrandt who throughout the twenties was the most powerful woman in the country Perhaps most surprising of all is Okrent s account of Joseph P Kennedy s legendary and long misunderstood role in the liquor business It s a book rich with stories from nearly all parts of the country Okrent s narrative runs through smoky Manhattan speakeasies where relations between the sexes were changed forever California vineyards busily producing sacramental wine New England fishing communities that gave up fishing for the more lucrative rum running business and in Washington the halls of Congress itself where politicians who had voted for Prohibition drank openly and without apology Last Call is capacious meticulous and thrillingly told It stands as the most complete history of Prohibition ever written and confirms Daniel Okrent s rank as a major American writer New England Journal of Education ,1921 **Mining and Scientific Press** ,1909

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