





Chapter - 2

Acid, Bases And Salts

ACIDS:

- These are the substances which have sour taste.
- They turn blue litmus solution red.
- They give H* ions in aqueous solution.
- The term 'acid' has been derived from the Latin word, acidus, which means sour.

Strong Acids: HCl, H₂SO₄, HNO₃

Weak Acids: CH3COOH, Oxalic acid, Lactic acid

Concentrated Acid: Having more amount of acid + less amount of water

Dilute Acid: Having more amount of water + less amount of acid

BASES:

- These are the substances which are bitter in taste and soapy in touch.
- They turn red litmus solution blue.
- They give OH⁻ ions in aqueous solution.

Strong Bases: NaOH, KOH, Ca(OH)₂

Weak Bases: NH₄OH

Alkalis: These are bases which are soluble in water [NaOH, KOH, Ca(OH)2].

SALTS:

These are the compounds formed from reaction of acid and base.





Topic 10 Acids Bases And Salts

Robert R. Redfield

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