

General Chemistry placement test Study Guide 2024 Solved 100%

Atomic Number - The number of protons in the nucleus of an atom. The atomic number of an element never changes, meaning that the number of protons in the nucleus of every atom in an element is always the same.

Hydrogen = 1, Carbon = 6, Oxygen = 8 (located on top of element of periodic table)

Mass Number - # of neutrons + # of protons = mass number (located on bottom of periodic table)

Isotopes - Let's say an atom is missing a neutron or has an extra neutron. That type of atom is called an isotope. An atom is still the same element if it is missing an electron.

Example: (carbon-14)

Distribution &/W subatomic particles in an Atom - A typical atom consists of three subatomic particles: Protons, Neutrons, and Electrons.

- Nucleons include protons and neutrons. All the positive charge of an atom is contained in the nucleus, and originates from the protons. Neutrons are neutrally-charged. Electrons, which are negatively-charged, are located outside of the nucleus.

Roles of subatomic particles: Protons, Neutrons, & Electrons - **Particle Mass:**

Protons & Neutrons have about same mass. Electrons have much less mass than protons and neutrons.

Particle Charge:

Protons have a positive charge (+)

- Neutrons have no charge (0)

- Net charge of the nucleus is positive and equal to the number of protons.

Usc Chemistry Placement Test Review

**College Research Group of Concord,
Massachusetts**



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Congressional Record United States. Congress,1992 **Resources in Education** ,1995-12 **Commerce Business Daily** ,2001 **Federal Register** ,1973-03 Scientific and Technical Aerospace Reports ,1977 **Naval Air Station, Alameda and Naval Supply Center, Oakland, U.S. Navy Military Construction Projects, Proposed New Dredging, San Francisco Bay** ,1990 **Highway Safety Literature** ,1977 *Who's who in California* ,1983 ROTC College Handbook, 1989-90 College Research Group of Concord, Massachusetts,1988 In depth profiles of all 560 colleges and universities offering ROTC or ROTC cross registration Covers all Army Navy and Air Force ROTC programs The College Handbook College Entrance Examination Board,1999 Presents information on 4 year colleges and universities and 2 year community colleges and technical schools *Chemical Engineering Progress* ,1961 **Who's Who in California** Sarah A. Vitale,1983-06 **Comparative Guide to American Colleges** James Cass,Max Birnbaum,1991-08 *SPE Reservoir Engineering* ,1994 **Who's Who in America** Marquis Who's Who, Inc,2002 *Who's who of Emerging Leaders in America* ,1989 *The Big Book of Colleges, 1997* Edward Custard,Princeton Review (Firm),1996-07-30 Lists more than 1 600 colleges and universities and provides information about admissions and academic programs *Who's who in American Education* ,1992 *Dissertation Abstracts International* ,1970 Resources in Education ,1979

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