

43. Electropositivity of elements across the periodic table normally
- remains constant down the group.
 - increases across the period.
 - decreases across the period.
 - decreases down the group.

44. Consider the following table:

Substance	Melting point / °C	Boiling point / °C
P	-78	-25
Q	-8	40
R	-6	150
S	44	289

Which of the substances is a liquid at room temperature and rapidly evaporates on exposure to air?

- P
 - Q
 - R
 - S
45. Which of the following oxides is amphoteric?
- Carbon(II) oxide
 - Nitrogen(IV) oxide
 - Lead(II) oxide
 - Calcium oxide
46. Which of the following processes occur during fractional distillation of petroleum?
- Condensation and diffusion
 - Diffusion and evaporation
 - Diffusion and sublimation
 - Evaporation and condensation
47. If the molar mass of $X(\text{HCO}_3)_2$ is 162 g mol^{-1} , determine the relative atomic mass of X. [H = 1.0, C = 12.0, O = 16.0]

- 40
- 48
- 61
- 101

48. Charcoal is used in the decolourization of sugar because of its

- absorption property.
- amorphous property.
- oxidizing property.
- adsorption property.

49. The first definition of an element was made by

- J. Dalton.
- A. Lavoisier.
- R. Boyle.
- J. J. Thompson.

50. Which of the following scientists formulated the law of conservation of mass?

- A. Lavoisier
- J. Dalton
- R. Boyle
- J. Proust

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Science Chemistry ,2024 Chemistry Phillip John Buddle,J. S. Mackay,1970

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