

Chemistry

Time: 2 hours

Marks: 60

- Five isomeric para-disubstituted aromatic compounds A to E with molecular formula $C_8H_8O_2$ were given for identification. Based on the following observations, give structures of the compounds.
 - Both A and B form a silver mirror with Tollen's reagent; also B gives a positive test with $FeCl_3$ solution.
 - C gives positive iodoform test.
 - D is readily extracted in aqueous $NaHCO_3$ solution.
 - E on acid hydrolysis gives 1,4-dihydroxybenzene.[5]
- 500 mL of 0.2 M aqueous solution of acetic acid is mixed with 500 mL of 0.2 M HCl at $25^\circ C$.
 - Calculate the degree of dissociation of acetic acid in the resulting solution and pH of the solution.
 - If 6 g of NaOH is added to the above solution, determine the final pH. [Assume there is no change in volume on mixing; K_a of acetic acid is $1.75 \times 10^{-5} \text{ mol L}^{-1}$].[5]
- Deduce the structures of $[NiCl_4]^{2-}$ and $[Ni(CN)_4]^{2-}$ considering the hybridization of the metal ion. Calculate the magnetic moment (spin only) of the species. [5]
- The density of the vapour of a substance at 1 atm pressure and 500 K is 0.36 kg m^{-3} . The vapour effuses through a small hole at a rate of 1.33 times faster than oxygen under the same condition.
 - Determine (i) molecular weight, (ii) molar volume, (iii) compression factor (Z) of the vapour and (iv) which forces among the gas molecules are dominating, the attractive or the repulsive?
 - If the vapour behaves ideally at 1000 K, determine the average translational kinetic energy of a molecule.[5]
- Write balance equations for the reactions of the following compounds with water;
 - Al_2C_3
 - CaNCN
 - BF_3
 - NCl_3
 - XeF_4[5]
- Identify X, Y and Z in the following synthetic scheme and write their structures.
$$CH_3CH_2C \equiv C-H \xrightarrow[(II) CH_3CH_2Br]{(I) NaNH_2} X \xrightarrow{H_2/Pd-BaSO_4} Y \xrightarrow{\text{alkaline } KMnO_4} Z$$
Is the compound Z optically active? Justify your answer. [5]
- How is boron obtained from borax? Give chemical equations with reaction conditions. Write the structure of B_2H_6 and its reaction with HCl.

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School Science Practical Work in Africa Umesh Ramnarain,2020-06-01 School Science Practical Work in Africa presents the scope of research and practice of science practical work in African schools It brings together prominent science educators and researchers from Africa to share their experience and findings on pedagogical innovations and research informed practices on school science practical work The book highlights trends and patterns in the enactment and role of practical work across African countries Practical work is regarded as intrinsic to science teaching and learning and the form of practical work that is strongly advocated is inquiry based learning which signals a definite paradigm shift from the traditional teacher dominated to a learner centered approach The book provides empirical research on approaches to practical work contextual factors in the enactment of practical work and professional development in teaching practical work This book will be of great interest to academics researchers and post graduate students in the fields of science education and educational policy *Proceedings of the ... Annual Conference and ... Conference of CASTME Africa* Science Teachers Association of Nigeria. Conference,2004

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