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 $2\text{ }3\text{Ca(OH)}_2 + 1\text{ }Al_2\text{SO}_4 \rightarrow 3\text{ }CaSO_4 + 2\text{ }Al(OH)_3$ type of reaction double displacement
 $3\text{ }3\text{Mg} + 1\text{ }Fe_2O_3 \rightarrow 2\text{ }Fe + 3\text{ }MgO$ type of reaction single displacement
 $4\text{ }1\text{ }C_2H_4 + 3\text{ }O_2 \rightarrow 2\text{ }CO_2 + 2\text{ }H_2O$

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 $NaOH + KNO_3 \rightarrow NaNO_3 + KOH$
 $CH_2O + CO_2 + H_2 \rightarrow 2\text{ }H_2O + 2\text{ }CO$
 $2\text{ }Fe + 6\text{ }NaBr \rightarrow 2\text{ }FeBr_3 + 6\text{ }Na$
 $3\text{ }Mg + 2\text{ }Fe_2O_3 \rightarrow 4\text{ }Fe + 3\text{ }MgO$
 $4\text{ }2\text{ }NH_4OH + 4\text{ }HBr \rightarrow 4\text{ }H_2O + 4\text{ }NH_4Br$
 $PbO + 2\text{ }Na_2CO_3 \rightarrow 2\text{ }Na_2O + CO_2 + 2\text{ }Pb$
 $4\text{ }H_2O + 8\text{ }Fe + 4\text{ }CO_2 \rightarrow 2\text{ }2\text{ }H_2O + 2\text{ }HCl + NaOH + 2\text{ }NaCl$

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