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P525/1
CHEMISTRY
(Theory)
Paper 1
Nov. /Dec. 2019.
2¾ hours.



UGANDA NATIONAL EXAMINATIONS BOARD

Uganda Advanced Certificate of Education

CHEMISTRY

(THEORY)

Paper 1

2 hours 45 minutes

INSTRUCTIONS TO CANDIDATES:

Answer **all** questions in Section A and **six** questions in Section B.

All questions **must** be answered in the spaces provided.

The Periodic Table, with relative atomic masses, is supplied.

Mathematical tables (3-figure tables) are adequate or non-programmable scientific electronic calculators may be used.

Illustrate your answers with equations where applicable.

Where necessary use the following:

Molar gas constant, R = $8.31 \text{ JK}^{-1} \text{ mol}^{-1}$

Molar volume of gas at s.t.p. is 22.4 litres

Standard temperature = 273 K

Standard pressure = 101325 Nm^{-2}

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1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	Total

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N Colangelo



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