

Science Study Guide: The Sun & Stars

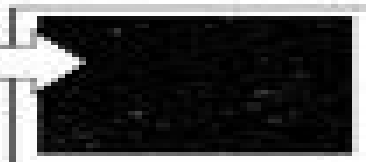


Lesson 1: Stars

Words To Know

1. Star - a giant ball of hot glowing gases
2. Light year - the distance light travels in one year

stars



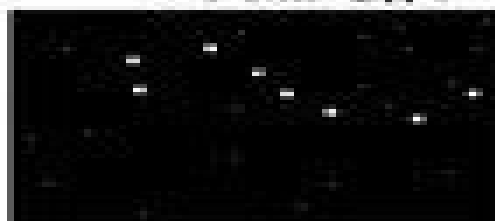
Applying Concepts

3. Light from the sun takes 8.3 minutes to reach Earth
4. Some stars look small because they are far away from Earth
5. Stars don't disappear during the day: the sun is just so bright you can't see them
6. Scientists describe stars by their brightness, color, size, and temperature
7. Stars can be red, orange, yellow, blue, or white
8. A star's color depends on its temperature
9. Blue stars are hotter than red or orange stars
10. Star patterns change with the different seasons
11. Sometimes stars make groups or patterns called constellations

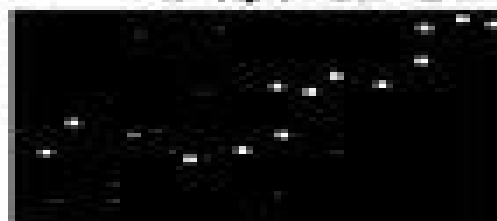
12. Big Dipper



13. Little Dipper



14. Hydra the Snake

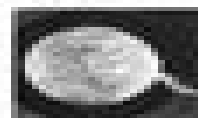


Be able to identify the constellations above

Lesson 2: The Sun

Words To Know

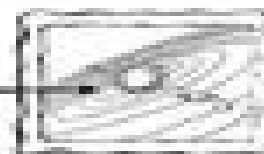
15. Sun - the closest star to Earth
16. Solar cycles - when the sun releases different amounts of energy in cycles
17. Solar flare - a sun storm that occurs when there is a burst of energy released from the sun
18. Sun spots - dark spots on the sun that are cooler than the area around them



Solar Flare

Applying Concepts

19. The sun is located at the center of the solar system
20. The surface of the sun is about 10,000°F (ten thousand degrees Fahrenheit)
21. The sun is NOT the biggest star. The Orbits is the biggest.
22. The sun appears to be so big and bright because it is the closest star to Earth.
23. It takes Earth 365 days just to go around/orbit around the sun.
24. The sun's energy travels to Earth in the form of light, heat, and food plants.
25. Solar flares can disrupt communication satellites and cause blackouts on Earth.



Stars Study Guide

Edgar Gardner Murphy



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web what is a titration a titration is an analytical procedure used to determine the accurate concentration of a sample by reacting it with a standard solution one type of titration uses a neutralization reaction in which an acid and a base react to produce a salt and water

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web donor whereas a base generically represented by b is a proton acceptor when acids and bases combine they swap a proton ha b a hb strong acids eg sulfuric acid have a very pronounced tendency to transfer a proton to a base whereas weak acids

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web acid base titrations laboratory report name rj hamdan date 11 8 section 432 ta ben kies a goals purpose of experiment 2 points 50 words the goal of this experiment is using the skills of precision to determine the

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web understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web oct 24 2022 acid base titrations lab report links file submission this page describes what you need to do to complete and submit the assignment

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web jul 19 2023 at the beginning of the titration the solution in the erlenmeyer flask is acidic as the base is added it completely reacts with the acid and the solution in the erlenmeyer flask continues to be acidic but at the equivalence point

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web answer based on the data the strong acid strong base combination of hydrochloric acid and sodium hydroxide had the neutral pH at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point

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web oct 1 2019 abstract titration is an analytical method used in biomedical sciences and analytical chemistry laboratories to determine the quantity or the concentration of a known or unknown substance there

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web may 10 2021 acid base titration lab dataclassroom a titration is a lab technique used to determine the volume of a solution that is needed to react with a given amount of another substance in this activity your goal is to determine the molar concentration and strength of two acid solutions by conducting titrations with a base solution of known

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web acid base titrations are lab procedures used to determine the concentration of a solution one of the standard laboratory exercises in general chemistry is an acid base titration during an acid base

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web spreadsheet computations prelab report this is the type of computations the computer does to determine first and second derivatives refer to these spreadsheet columns for the comments below pH titration graph lab data provide columns a

and b

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web 1 approximately 3 g of hydrated oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ was weighted to the nearest 0.01 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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web aug 30 2022 acid base titrations are usually used to find the amount of a known acidic or basic substance through acid base reactions the analyte titrand is the solution with an unknown molarity

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web this problem has been solved you'll get a detailed solution from a subject matter expert that helps you learn core concepts see answer question part b acid base titration lab report i determining the concentration of sulfuric acid solution H_2SO_4 by titrating it with NaOH solution

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web solution a titrant volume 0 ml the solution pH is due to the acid ionization of HCl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0.100 M the pH of the solution is then $\text{pH} = -\log 0.100 = 1.000$ pH log 0.100 = 1.000 b titrant volume 12.50 ml

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web fundamental chemistry i chem 101 pre lab 1 pre lab report for the first experiment post lab 1 post lab report for the first experiment lab 5 acid base titration determining the concentration of a NaOH solution lab partner n a acid base titration determining the concentration of a NaOH solution

titration screen experiment resource rsc education - Mar 20 2022

web rp 1 make up a volumetric solution and carry out a simple acid base titration physical chemistry acids and bases pH curves titrations and indicators titrations of acids with bases students should be able to perform calculations for these titrations based on experimental results inorganic chemistry transition metals variable

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web one type of titration uses a neutralization reaction in which an acid and a base react to produce a salt and water in equation 1 the acid is HCl hydrochloric acid and the base is NaOH sodium hydroxide when the acid and base react they form NaCl sodium chloride which is also known as table salt

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web acid and base titrations lab report chm 114 jx abstract this goal was to give us experience finding the standardization of

through the use of a primary standard in this experiment we will be using NaOH and HCl as well as KHP

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web question lab report acid base titrations purpose of the experiment in this experiment we will titrate an acid with a base to determine the concentration of an acid with a base to determine the concentration of an acidic sample