

Unit 2 Worksheet: St. Michael's Concentration

Part A: Describing Matter Concentration

1. Sodium chloride has a chemical formula of NaCl . Explain what this concentration tells you about this NaCl solution. What might a chemical name like this tell you?
2. A 100 mL sample of potassium nitrate is dissolved in enough water to make 200 mL of solution. Calculate the mass percent of the potassium nitrate.
3. Find the molarity of a solution made from 200 g of CaCl_2 dissolved in enough water to make 1.0 L.
4. An unknown liquid contains sodium hydroxide and is labeled as containing 40% sodium hydroxide. A 100 g sample of the liquid is dissolved in enough water to make 200 mL of solution. Calculate the molarity of sodium hydroxide in this solution.
5. What mass percent of sodium hydroxide is present in a 100 g sample of a solution made from 10 g of NaOH and 90 g of water?

Unit 9 Worksheet 3 Molar Concentration Part B

CO Houle



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