

Unit 1 Chemistry Study Guide

History of the Atom:

Democritus (450 BCE):

- Matter is not infinitely divisible.
- There are single particles that cannot be divided further, called **atomos**.

Aristotle (380 BCE):

- Matter is infinitely divisible.
- Matter is called **hyle**.

Antoine Lavoisier (1774):

- **Law of Conservation of Mass:** Mass cannot be created or destroyed.

Joseph Proust (1779):

- **Law of Definite Proportions:** All samples of a pure compound contain the same elements in the same proportion by mass (no matter quantity).

John Dalton (1803 – 1808):

- Observed work of Lavoisier and Proust
- Led Dalton to develop the first **atomic theory**.
- **Dalton's Atomic Theory:**
 - o All matter is composed of extremely small particles called atoms.
 - o All atoms of a given element are identical, having the same size, mass, and chemical properties. The atoms of one element are different from the atoms of all other elements.
 - o Atoms cannot be subdivided, created, or destroyed.
 - o Atoms of different elements can combine in simple whole number ratios to form chemical compounds.
 - o In chemical reactions, atoms are combined, separated or rearranged.
- **Law of Multiple Proportions:** Elements can combine in different proportions. However, the different compounds will have different properties but the ratio of atoms is always a whole number.

Billiard Ball Model:



atom

J.J. Thompson (1906):

- **Cathode Ray Experiment –**
 - o Part 1: Thompson ran electricity through a cathode ray tube and when he put the negatively charged end of a magnet near the tube, he saw the ray of electricity

Unit Chemistry Study Guide Answers

Siddappa N.Byrareddy



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Adopting the Melody of Expression: An Psychological Symphony within **Unit Chemistry Study Guide Answers**

In a world taken by monitors and the ceaseless chatter of instant conversation, the melodic beauty and psychological symphony produced by the prepared term usually disappear in to the backdrop, eclipsed by the constant noise and disruptions that permeate our lives. However, situated within the pages of **Unit Chemistry Study Guide Answers** a stunning literary treasure overflowing with fresh thoughts, lies an immersive symphony waiting to be embraced. Crafted by a masterful composer of language, this interesting masterpiece conducts readers on a psychological trip, skillfully unraveling the concealed songs and profound influence resonating within each cautiously constructed phrase. Within the depths of this touching review, we will explore the book is central harmonies, analyze their enthralling writing design, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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