

Index Number:

Answer all the questions.

1. In a volumetric analysis, 24.80 cm^3 of $0.100 \text{ mol dm}^{-3}$ of a mineral acid, Y, neutralized 25.0 cm^3 of an alkaline solution containing 5.80 g of Na_2CO_3 per dm^3 .

(a) From the information provided, calculate the:

- (i) number of moles of Y;

- (ii) number of moles of Na_2CO_3 ;

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- (iii) mole ratio of the acid to base in the analysis.

ZAMGIST

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