# Answer all the questions.

- In a volumetric analysis, 24.80 cm<sup>3</sup> of 0.100 mol dm<sup>-3</sup> of a mineral acid, Y, neutralized 25.0 cm<sup>3</sup> of an alkaline solution containing 5.80 g of Na<sub>2</sub>CO<sub>3</sub> per dm<sup>3</sup>.
  - (a) From the information provided, calculate the:
    - (i) number of moles of Y:

(ii) number of moles of Na<sub>2</sub>CO<sub>3</sub>;

(iii) mole ratio of the acid to base in the analysis.

STEAM

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