

## QUESTION 12 (contd.)

- (c)
- Decay constant
  - Number of particles present.

(d) Given: Activity,  $A = 2.0 \times 10^6$  Bq or decay/s  
half life,  $T_{1/2} = 1.8 \times 10^8$  secs.

Now,

$$\text{decay constant, } \lambda = \frac{\ln 2}{T_{1/2}} = \frac{\ln 2}{1.8 \times 10^8}$$

$$A = \lambda N$$

$$\text{Number of isotopes in source} = \frac{A}{\lambda} = \frac{2.0 \times 10^6}{\left( \frac{\ln 2}{1.8 \times 10^8} \right)}$$

$$N = 5.19 \times 10^{14} \text{ nuclei}$$

# Waec2014 Physics Obj And Essay

**Mavis Kitcher (Mrs)**



**Waec2014 Physics Obj And Essay:**

**Junior Graphic** Mavis Kitcher (Mrs),2014-02-26

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