

Chapter 22 Solutions

Problem 1:

A $+1.5\text{ microC}$ charge is located 40 cm from a -3.0 microC charge. The magnitude of the electrostatic force on the larger charge and on the smaller charge (in N) is, respectively.

Answer: 2.5, 2.5

Solution: The magnitude of the electrostatic force is given by,

$$F = \frac{KQ_1Q_2}{r^2} = \frac{(8.99 \times 10^9 \text{ Nm}^2/\text{C}^2)(1.5\mu\text{C})(3.0\mu\text{C})}{(40\text{cm})^2} = 2.53\text{N}$$

Remember the force on 1 due to 2 is equal and opposite to the force on 2 due to 1.

Problem 2:

Two point particles have charges q_1 and q_2 and are separated by a distance d . Particle q_2 experiences an electrostatic force of 1.2 mN due to particle q_1 . If the charges of both particles are doubled and if the distance between them is doubled, what is the magnitude of the electrostatic force between them (in mN)?

Answer: 0.3

Solution: The magnitude of the initial electrostatic force is

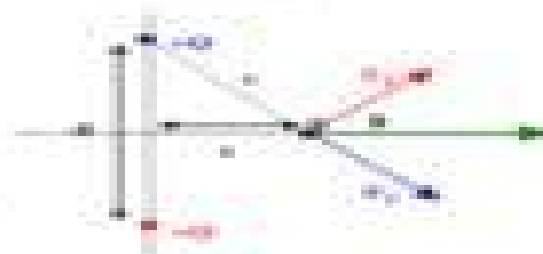
$$F_i = \frac{Kq_1q_2}{d^2} = 1.2\text{mN}$$

The magnitude of the final electrostatic force is

$$F_f = \frac{K(2q_1)(2q_2)}{(2d)^2} = F_i = 1.2\text{mN}$$

Problem 3:

Two identical point charges $+Q$ are located on the y -axis at $y=+d/2$ and $y=-d/2$, as shown in the figure. A third charge q is placed on the x -axis. At what distance from the origin is the net force on q a maximum?



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Electrochemical Methods Allen J. Bard, Larry R. Faulkner, Henry S. White, 2022-05-03 The latest edition of a classic textbook in electrochemistry The third edition of *Electrochemical Methods* has been extensively revised to reflect the evolution of electrochemistry over the past two decades highlighting significant developments in the understanding of electrochemical phenomena and emerging experimental tools while extending the book's value as a general introduction to electrochemical methods This authoritative resource for new students and practitioners provides must have information crucial to a successful career in research The authors focus on methods that are extensively practiced and on phenomenological questions of current concern This latest edition of *Electrochemical Methods* contains numerous problems and chemical examples with illustrations that serve to illuminate the concepts contained within in a way that will assist both student and mid career practitioner Significant updates and new content in this third edition include An extensively revised introductory chapter on electrode processes designed for new readers coming into electrochemistry from diverse backgrounds New chapters on steady state voltammetry at ultramicroelectrodes inner sphere electrode reactions and electrocatalysis and single particle electrochemistry Extensive treatment of Marcus kinetics as applied to electrode reactions a more detailed introduction to migration and expanded coverage of electrochemical impedance spectroscopy The inclusion of Lab Notes in many chapters to help newcomers with the transition from concept to practice in the laboratory The new edition has been revised to address a broader audience of scientists and engineers designed to be accessible to readers with a basic foundation in university chemistry physics and mathematics It is a self contained volume developing all key ideas from the fundamental principles of chemistry and physics Perfect for senior undergraduate and graduate students taking courses in electrochemistry physical and analytical chemistry this is also an indispensable resource for researchers and practitioners working in fields including electrochemistry and electrochemical engineering energy storage and conversion analytical chemistry and sensors

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