

CHEMISTRY

NEW!

Unit 13 Thermochemistry Learning Activities & Tests

13.1 Entropy of Reaction: Bond Energies

Calculate the enthalpy of reaction for a reaction using bond energies.

Worksheet: The enthalpy of reaction can be calculated using bond energies. Calculate the enthalpy of reaction for the following reaction.

Reaction: $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$



For this reaction, calculate the enthalpy of reaction using bond energies.



Unit 13 Calculating: Entropy of Reaction

Calculate the enthalpy of reaction for a reaction using bond energies. Calculate the enthalpy of reaction for the following reaction.

Reaction: $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$

Worksheet: The enthalpy of reaction can be calculated using bond energies. Calculate the enthalpy of reaction for the following reaction.

1. Calculate the enthalpy of reaction for the reaction $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$.
2. Calculate the enthalpy of reaction for the reaction $\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$.

13.2 Heating and Cooling Curves

Interpret a heating and cooling curve. Calculate the enthalpy of reaction for a reaction using bond energies. Calculate the enthalpy of reaction for the following reaction.

Reaction: $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$

Worksheet: The enthalpy of reaction can be calculated using bond energies. Calculate the enthalpy of reaction for the following reaction.

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Worksheet: The enthalpy of reaction can be calculated using bond energies. Calculate the enthalpy of reaction for the following reaction.

Reaction: $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$

13.3 Energy Diagrams

Interpret a heating and cooling curve. Calculate the enthalpy of reaction for a reaction using bond energies. Calculate the enthalpy of reaction for the following reaction.

1. Calculate the enthalpy of reaction for the reaction $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$.
2. Calculate the enthalpy of reaction for the reaction $\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$.



Worksheet: The enthalpy of reaction can be calculated using bond energies. Calculate the enthalpy of reaction for the following reaction.



Unit 13 Thermochemistry Test Review

Arturo Cuomo



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