

Chemistry – Unit 7 Worksheet 3

Write balanced chemical equations for the following reactions.

Note: All coefficients of 1 are omitted; in the first 10 items, the coefficients are all 1.

1. Ammonia (NH_3) reacts with hydrogen chloride to form ammonium chloride.
 $\text{NH}_3 + \text{HCl} \rightarrow \text{NH}_4\text{Cl}$

2. Calcium carbonate decomposes upon heating to form calcium oxide and carbon dioxide.
 $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$

3. Barium oxide reacts with water to form barium hydroxide.
 $\text{BaO} + \text{H}_2\text{O} \rightarrow \text{Ba(OH)}_2$

4. Acetaldehyde (CH_3CHO) decomposes to form methane (CH_4) and carbon monoxide.
 $\text{CH}_3\text{CHO} \rightarrow \text{CH}_4 + \text{CO}$

5. Zinc reacts with copper (II) nitrate to form zinc nitrate and copper.
 $\text{Zn} + \text{Cu(NO}_3)_2 \rightarrow \text{Zn(NO}_3)_2 + \text{Cu}$

6. Calcium sulfite decomposes when heated to form calcium oxide and sulfur dioxide.
 $\text{CaSO}_3 \rightarrow \text{CaO} + \text{SO}_2$

7. Iron reacts with sulfuric acid (H_2SO_4) to form iron (II) sulfate and hydrogen gas.
 $\text{Fe} + \text{H}_2\text{SO}_4 \rightarrow \text{FeSO}_4 + \text{H}_2$

8. A nitrogen containing carbon compound, $\text{C}_2\text{H}_6\text{N}_2$, decomposes to form ethane, C_2H_6 , and nitrogen gas.
 $\text{C}_2\text{H}_6\text{N}_2 \rightarrow \text{C}_2\text{H}_6 + \text{N}_2$

9. Phosgene, COCl_2 , is formed when carbon monoxide reacts with chlorine gas.
 $\text{CO} + \text{Cl}_2 \rightarrow \text{COCl}_2$

Unit 7 Ws 3 Modeling Chemistry

**Ryzhard Pohorecki, John Bridgwater, M.
Molzahn. Rafiqul Gani and Crispulo
Gallegos**

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Monthly Catalogue, United States Public Documents, 1989-05 *Monthly Catalog of United States Government Publications*, 2002-07 *Popular Photography*, 1982-03 Tenth International Symposium on Chemical Reaction Engineering J. R. Bourne, W. Regenass, W. Richarz, 2017-05-04 ISCRE 10 Tenth International Symposium on Chemical Reaction Engineering documents the proceedings of the symposium which brought together experts from all over the world to discuss developments in CRE Efforts were made to cover high added value substances and to encourage papers from industry Some

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