

UNIT VII Review Stoichiometry

Stoichiometry Calculations involving MASS, MOLES, VOLUME, and MOLECULES

1. Consider the reaction $4\text{NH}_3 + 5\text{O}_2 \rightarrow 6\text{H}_2\text{O} + 4\text{NO}$
 - a. What mass of NO is produced when 5.00 mol of NH_3 are reacted with excess O_2 ? 150g NO
 - b. What volume of O_2 at STP is required to produce 90.0L of NO? 113L O_2
 - c. How many molecules of H_2O are formed when 100.0g of NH_3 react? 5.31×10^{24} molecules H_2O
2. Consider the reaction $\text{SiCl}_4 + 2\text{H}_2 \rightarrow \text{Si} + 4\text{HCl}$
 - a. How many moles of SiCl_4 are required to produce 4.25×10^{12} molecules of HCl. 1.76×10^{-12} mol SiCl_4
 - b. If 25.0 mL of H_2 (g) at STP react, calculate the mass of Si produced. 0.0157g Si
 - c. If 0.892 kg of SiCl_4 react, calculate the number of molecules of HCl produced. 21.0 mol HCl
3. Silver reacts with nitric acid, HNO_3 to form nitrogen monoxide, NO, silver nitrate, AgNO_3 and water.

O M I T

 - a. How many grams of nitric acid are required to react with 5.00g of silver?
 - b. How many grams of silver nitrate will be formed?
 - c. How many moles of nitrogen monoxide will be produced?

Stoichiometry using MOLAR CONCENTRATION

4. Consider the following reaction $\text{CaCO}_3 + 2\text{HCl} \rightarrow \text{CaCl}_2 + \text{CO}_2 + \text{H}_2\text{O}$
 - a. What volume of HCl having a concentration of 0.0010M is neutralized by a 0.850 g portion of CaCO_3 ? 17.0L HCl
 - b. What volume of CO_2 at STP is produced if 1.25L of 0.0055 M HCl reacts with an excess of CaCO_3 ? 0.077L CO_2
 - c. How many moles of 0.200 M HCl is reacted to produce 3.00g of CaCl_2 ? 0.024 mol H_2
5. If 17.5 g of zinc are reacted with phosphoric acid, H_3PO_4 , then zinc phosphate, $\text{Zn}_3(\text{PO}_4)_2$, and hydrogen are produced.
 - a. How many moles of phosphoric acid are required? 0.178 mol H_3PO_4
 - b. If the phosphoric acid is a 3.00M solution, how many litres of the solution are needed? 0.0593L
 - c. What mass of zinc phosphate will be produced? 84.4g $\text{Zn}_3(\text{PO}_4)_2$

Unit Vii Review Key

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Phonics Book 3 Kathy Dickerson Crane, 2004-03-23 This three book series helps students develop reading skills Students review and practice phonemic awareness and alphabet recognition skills and work their way through more advanced phonics concepts including blends digraphs synonyms and antonyms Suitable for ELL and

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D. Renfrow, *The Teacher's Handbook* serves as an aid in curriculum development and daily lesson planning. Suggested daily lesson plans, suggested assignments following each lesson plan, teaching tips for each unit, suggested examinations for the semester and answer keys for the written exercises and unit review worksheets. It also suggests ways to successfully integrate keyboard and computer technology into the curriculum. **Interactive Learning: Daily Sentence Editing Grd 1** Teacher Created Resources, 2011. Use these fully interactive products to teach students important skills in grammar, punctuation and spelling. The self-correcting activities work on ALL brands of interactive whiteboards. The PC/Mac CD provides 180 to 270 ready-to-edit sentences per grade. In addition, thousands more sentences and paragraphs can be created and saved. The book format makes it easy for teachers to see the entire scope of the product at a glance and to copy pages as needed for individual work.

Changes 3 Teacher's Book Jack C. Richards, Jonathan Hull, Susan Proctor, 1996-07-18. *Changes* is a three-level general English course for adult and young adult learners. *Changes* ensures that students have every opportunity to develop confident communicative ability as well as accuracy in English.

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