

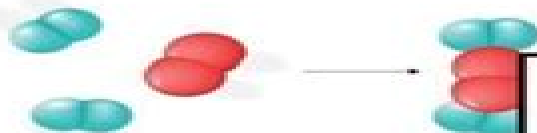
Rate of Reaction & Equilibrium Unit Bundle

Rates of reaction

A rate is a measure of how much something changes over a given period of time. In chemistry, the rate of a reaction can be defined as how quickly reactants are lost or how quickly a product is formed. If the concentration of a reactant lost or product formed is being measured, the unit will be moles per liter per unit of time [M/s, M/min, M/hour]. If you're measuring the amount of gas formed during a reaction the units will be a unit of mass per unit of time [g/s, g/min, g/hour]. The units for a change in the amount of gas are typically a unit of volume per unit of time [c Litres, etc].

A chemical reaction occurs when two or more particles react with one another to form a new product. According to collision theory, there are three main factors that must occur in order for a chemical reaction to occur:

- 1.) Substances must collide with one another.
- 2.) They must collide with the correct orientation.
- 3.) The substances must collide with sufficient energy for the reaction to take place. The minimum energy that colliding substances need in order to react is known as the activation energy.



For a given set of conditions, every chemical reaction occurs. The rate of reaction is typically fastest at the maximum concentration of reactants that can collide. As the reaction goes on more product is being produced and molecules remaining.

Some reactions are completed in seconds, while others can take years for them to be completed. If you want a chemical reaction, there are several factors that you can change: the concentration of reactants, temperature, catalyst, and in some cases, light.

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1 - Equilibrium



What does it mean if a reaction is in equilibrium?

Part 6: Short Answer

Read each question carefully and answer it fully.

101.) What is it mean if a reaction is in equilibrium?

102.) What does Le Chatelier's principle state?



Science Hedgehog

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Unit 16 Reaction Rates Notes Answers

Peipei Pang

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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