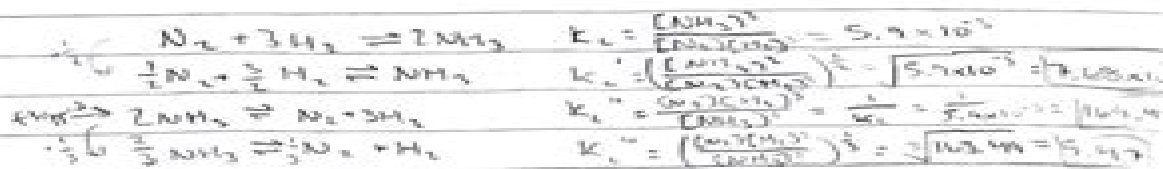


WS 4

49. $k'_c = 7.68 \times 10^2$
 $k''_c = 169.49$
 $k'''_c = 5.53 + 18$



2. $\sqrt{3872.62}$



$$b) 2.40 \times 10^8$$



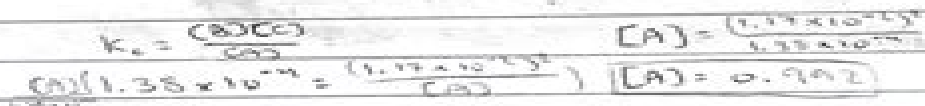
$$1.55 \times 10^{-4}$$



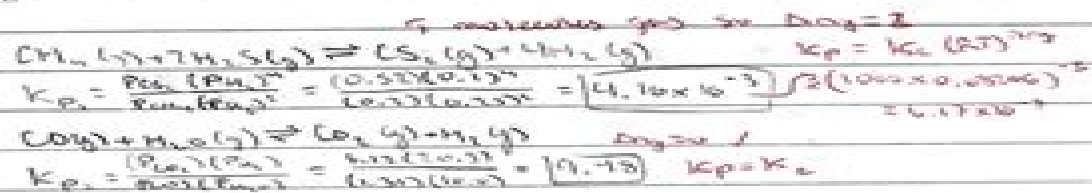
3. 1.38×10^{-4}



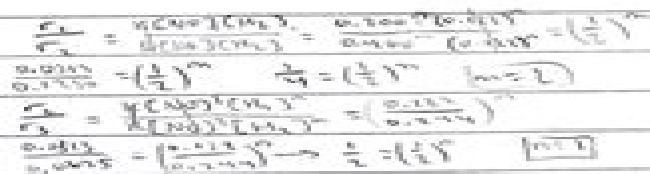
★ b) $P(A) = 0.982$



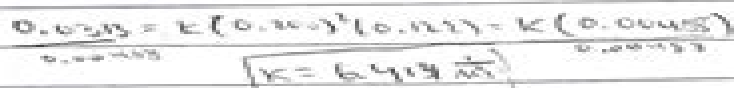
4. $K_{eq} = K_p = 4.16 \times 10^{-3}$



5. ~~a)~~ $\text{CaCl}_2 = \text{K}^+ [\text{NO}_3]^- (\text{M}_2)$



b) $\kappa = 6.414 \frac{1}{\text{nm}}$



④ c) 0.0866 Mgs



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