

Unit 08 – Stoichiometry – Worksheet 1

Answer the following questions in the space provided at the left.

- 18 mol O₂ 1. How many moles of oxygen are produced from twelve moles of Potassium chlorate?



- 21 mol O₂ 2. How many moles of oxygen are produced from 14 moles of Potassium chlorate?



- 45 mol H₂O 3. How many moles of water are produced from 22.5 moles of oxygen?



- 20 mol HCl 4. Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.



- 21 mol Cl₂ 5. Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.



Unit 8 Stoichiometry Test Practice

RS Peters



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SAT Subject Test Chemistry Joseph A. Mascetta, Mark Kernion, 2020-12-01 Note College Board has discontinued the SAT Subject Tests in the US The tests will be available outside the US in June 2021 and then be discontinued Barron's SAT Subject Test Chemistry with 7 Practice Tests features in depth review of all topics on the exam and full length practice tests in the book and online This edition includes One full length diagnostic test to help you assess your strengths and weaknesses Comprehensive review of all topics on the exam including introductory chemistry atomic structure and the periodic table bonding chemical formulas gases and laws stoichiometry liquids solids and phase changes chemical reactions and thermochemistry chemical reactions chemical equilibrium acids bases and salts oxidation reduction carbon and organic chemistry and the laboratory Four full length practice tests that reflect the actual SAT Subject Test Chemistry exam in length question types and degree of difficulty Two full length online practice tests with answer explanations and automated scoring Appendices which include the periodic table important equation constant and data tables and a glossary of chemistry terms

Reflection High-Energy Electron Diffraction and Reflection Electron Imaging of Surfaces P.K. Larsen, P.J. Dobson, 2012-12-06 This volume contains the papers presented at the NATO Advanced Research Workshop in Reflection High Energy Electron Diffraction and Reflection Electron Imaging of Surfaces held at the Koningshof conference center Veldhoven the Netherlands June 15 19 1987 The main topics of the workshop Reflection High Energy Electron Diffraction RHEED and Reflection Electron Microscopy REM have a common basis in the diffraction processes which high energy electrons undergo when they interact with solid surfaces at grazing angles However while REM is a new technique developed on the basis of recent advances in transmission electron microscopy RHEED is an old method in surface crystallography going back to the discovery of electron diffraction in 1927 by Davisson and Germer Until the development of ultra high vacuum techniques in the 1960's made instruments using slow electrons more accessible RHEED was the dominating electron diffraction technique Since then and until recently the method of Low Energy Electron Diffraction LEED largely surpassed RHEED in popularity in surface studies The two methods are closely related of course each with its own specific advantages The grazing angle geometry of RHEED has now become a very useful feature because this makes it ideally suited for combination with the thin growth technique of Molecular Beam Epitaxy MBE This combination allows in situ studies of freshly grown and even growing surfaces opening up new areas of research of both fundamental and technological importance

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