

Unit 9 Worksheet: St. Michael's Concentration

Part 9: Describing Molar Concentrations

1. Sodium chloride was dissolved in water to produce a 1.5M solution. Suppose that this same solution had been added to the flask solution. This would be incorrect. **Just this solution.**
Then, calculate the molarities of the two solutions. The molarity of the solution is 1.5M. The molarity of the solution is 1.5M.
2. A 10.0 g solution of potassium chloride is dissolved in enough water to make 100.0 mL of solution. Calculate the molar concentration of this potassium chloride.

$$10.0 \text{ g} \times \frac{1 \text{ mol}}{74.55 \text{ g}} = 0.134 \text{ mol}$$

$$\text{Molarity} = \frac{0.134 \text{ mol}}{0.100 \text{ L}} = 1.34 \text{ M}$$

3. Find the molarity of a solution made from 25.0 g of CaCl_2 dissolved in enough water to make 0.50 L.

$$25.0 \text{ g} \times \frac{1 \text{ mol}}{110.98 \text{ g}} = 0.225 \text{ mol}$$

$$\text{Molarity} = \frac{0.225 \text{ mol}}{0.50 \text{ L}} = 0.45 \text{ M}$$

4. An unknown ionic solution's "molarity" is defined by preparing a 1.00 g of solution sample and measuring the volume of solution in mL. Calculate the molarity of the solution in the solution.

$$1.00 \text{ g} \times \frac{1 \text{ mol}}{74.55 \text{ g}} = 0.0134 \text{ mol}$$

$$\text{Molarity} = \frac{0.0134 \text{ mol}}{0.100 \text{ L}} = 0.134 \text{ M}$$

5. A 10.0 g solution of potassium chloride is dissolved in enough water to make 100.0 mL of solution. Calculate the molar concentration of this potassium chloride.

$$10.0 \text{ g} \times \frac{1 \text{ mol}}{74.55 \text{ g}} = 0.134 \text{ mol} \quad \frac{0.134 \text{ mol}}{0.100 \text{ L}} = 1.34 \text{ M}$$

Unit 9 Worksheet 3 Molar Concentration Answer Key

Xiaolong Qi



Unit 9 Worksheet 3 Molar Concentration Answer Key:

Unveiling the Power of Verbal Artistry: An Psychological Sojourn through **Unit 9 Worksheet 3 Molar Concentration Answer Key**

In a world inundated with screens and the cacophony of immediate interaction, the profound energy and psychological resonance of verbal beauty often fade into obscurity, eclipsed by the constant onslaught of noise and distractions. However, located within the lyrical pages of **Unit 9 Worksheet 3 Molar Concentration Answer Key**, a captivating perform of literary brilliance that impulses with natural feelings, lies an wonderful trip waiting to be embarked upon. Composed by way of a virtuoso wordsmith, this mesmerizing opus manuals viewers on an emotional odyssey, delicately revealing the latent possible and profound impact stuck within the elaborate web of language. Within the heart-wrenching expanse with this evocative examination, we shall embark upon an introspective exploration of the book is central themes, dissect its charming publishing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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