

Section Three: Extended answer**40% (60 marks)**

This section contains **five (5)** questions. You must answer **all** questions. Write your answers in the spaces provided.

Where questions require an explanation and/or description, marks are awarded for the relevant chemical content and also for coherence and clarity of expression. Lists or dot points are unlikely to gain full marks.

Final answers to calculations should be expressed to the appropriate number of significant figures.

Spare pages are included at the end of this booklet. They can be used for planning your responses and/or as additional space if required to continue an answer.

- **Planning:** If you use the spare pages for planning, indicate this clearly at the top of the page.
- **Continuing an answer:** If you need to use the space to continue an answer, indicate in the original answer space where the answer is continued, i.e. give the page number. Fill in the number of the question that you are continuing to answer at the top of the page.

Suggested working time: 70 minutes.

Question 37**(16 marks)**

The Contact process is now the most commonly used method of manufacturing sulfuric acid (H_2SO_4). Sulfuric acid is a colourless, thick liquid with a slightly pungent odour. It is extremely corrosive and can cause severe damage upon skin contact. In some cases, sulfuric acid is dyed a brown colour during production, to alert people to the dangerous nature of the substance. Sulfuric acid is widely used in mineral processing and fertiliser production, as well as many other industries.

In the Contact process, sulfur is first combined with oxygen to form sulfur dioxide. The sulfur dioxide is then converted to sulfur trioxide in the reversible reaction (Step 1) shown below.



- (a) What conditions of pressure would be required to achieve the maximum yield and the maximum rate of formation of SO_3 in Step 1? (circle your answer) (2 marks)
- | | | | | |
|------|--|---------------|---|--------------|
| (i) | Maximum SO_3 yield: | high pressure | / | low pressure |
| (ii) | Maximum rate of SO_3 formation: | high pressure | / | low pressure |
- (b) Suggest why a low pressure, just above atmospheric, is used for Step 1. (1 mark)

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 - The Rise of Digital Reading Wace Examination Papers
 - Advantages of eBooks Over Traditional Books
2. Identifying Wace Examination Papers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Wace Examination Papers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Wace Examination Papers

- Personalized Recommendations
- Wace Examination Papers User Reviews and Ratings
- Wace Examination Papers and Bestseller Lists
- 5. Accessing Wace Examination Papers Free and Paid eBooks
 - Wace Examination Papers Public Domain eBooks
 - Wace Examination Papers eBook Subscription Services
 - Wace Examination Papers Budget-Friendly Options
- 6. Navigating Wace Examination Papers eBook Formats
 - ePub, PDF, MOBI, and More
 - Wace Examination Papers Compatibility with Devices
 - Wace Examination Papers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Wace Examination Papers
 - Highlighting and Note-Taking Wace Examination Papers
 - Interactive Elements Wace Examination Papers
- 8. Staying Engaged with Wace Examination Papers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Wace Examination Papers
- 9. Balancing eBooks and Physical Books Wace Examination Papers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Wace Examination Papers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Wace Examination Papers
 - Setting Reading Goals Wace Examination Papers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Wace Examination Papers

- Fact-Checking eBook Content of Wace Examination Papers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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