

10.1 Periodic Trends



10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1: Introduction

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 10. It exists in the periodic table.
- _____ 11. It has no line periodic table.
- _____ 12. Group 1 elements.
- _____ 13. Elements that are strong oxidizing agents.
- _____ 14. Group 18 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. noble gases

In the space on the left, write one of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 15. There are two representative elements of elements.
- _____ 16. Most elements have higher of the elements in the periodic table are metals.
- _____ 17. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 18. Group 1A elements are the most reactive metals.
- _____ 19. Group 1A elements are highly reactive nonmetals known as halogens.
- _____ 20. Group 1A elements are very reactive elements known as representative metals.
- _____ 21. Alkali metals have properties of both metals and nonmetals.

Study Guide For Content Mastery Chemistry Answer

**Juliette Lantz, Renée Cole, The POGIL
Project**



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Glencoe Chemistry Glencoe/McGraw-Hill,2002 Study Guide to Accompany Basics for Chemistry Martha Mackin,2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook McGraw Hill,2001-08 Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change **Chemistry** McGraw-Hill Staff,2001-03 *Chemistry in Action: Strategies and Techniques for Mastering the Advanced Placement Examination* Pasquale De Marco,2025-07-20 Chemistry in Action Strategies and Techniques for Mastering the Advanced Placement Examination is the ultimate resource for students preparing for the Advanced Placement Chemistry examination With its comprehensive coverage of topics engaging explanations and abundant practice opportunities this book sets students on the path to success Inside Chemistry in Action Strategies and Techniques for Mastering the Advanced Placement Examination readers will find a wealth of valuable content tailored to the AP Chemistry curriculum Clear and concise explanations of fundamental concepts are complemented by numerous examples illustrations and practice problems These elements work together to foster a deep understanding of chemical principles and prepare students for the rigors of the AP Chemistry exam Beyond its focus on exam preparation Chemistry in Action Strategies and Techniques for Mastering the Advanced Placement Examination also provides a broader perspective on the field of chemistry It delves into the practical applications of chemistry in various industries showcasing how chemical principles impact our everyday lives This approach not only enhances students appreciation for the subject but also equips

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