

Physical Science A

Final Exam Study Guide

- **Matter**
 - Compare and contrast physical and chemical properties, and physical and chemical changes
 - Compare and contrast properties and behaviors of atoms in each of the 3 states of matter
 - Convert between mass, volume, and density, and relate appropriate SI units for each variable
 - Predict how solutions of different densities would settle in a graduated cylinder
- **Atoms, Elements, and the Periodic Table**
 - Describe Antoine Lavoisier and Dmitri Mendeleev's contributions to the development of the modern periodic table
 - Define the term ion and be able to predict the most common ions formed for elements by looking at the periodic table
 - Identify three subatomic particles and compare their properties
 - Compare and contrast atoms in ground and excited states
 - Distinguish the atomic number of an element from the mass number of an isotope, and use these numbers to describe the structure of atoms
 - Describe the currently accepted model of an atom
 - Interpret the information provided about the atoms and elements in a balanced chemical reaction
 - Differentiate between elements that are good and poor conductors of heat and electricity
 - Understand the development and organization of the periodic table, especially in terms of group properties, atomic mass and valence electrons
 - Identify general properties of metals, nonmetals, and metalloids
 - Define the words malleable, brittle, volatile, and ductile
 - Explain how the electron cloud model represents the behavior and locations of electrons in atoms
 - Predict the reactivity of some elements based on their locations in the periodic table
 - Compare and contrast Bohr and Rutherford's atomic models
 - Distinguish the atomic number of an element from the mass number of an isotope, and use these numbers to describe the structure of atoms
- **Bonding**
 - List the monomers that make up polymers for different biological macromolecules
 - Compare and contrast ionic and covalent bonds
 - Understand the development and organization of the periodic table, especially in terms of group properties, atomic mass and valence electrons
 - Explain the role that valence electrons play in bonding
 - Explain the octet rule for bonding
 - Draw Lewis dot diagrams for elements, and use them to predict bonding between one or more atoms
 - Identify three subatomic particles and compare their properties
 - Compare and contrast atoms in ground and excited states
- **Chemical Reactions**
 - Describe factors that can be used to increase reaction rates
 - Classify chemical reactions as synthesis, decomposition, single-replacement, double-replacement, or combustion reactions
 - Predict the products of chemical reactions
 - Interpret chemical equations in terms of reactants, products, and conservation of mass
 - Balance chemical equations by manipulating coefficients
 - Calculate amounts of reactants or products by using molar mass, molar ratios, and balanced chemical equations
 - Apply the Law of Conservation of Mass to chemical reactions
 - List ways to tell that a chemical reaction has occurred
 - Distinguish between endothermic and exothermic reactions

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Richard Bailey



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web if all three pairs of corresponding sides are congruent the triangles are congruent this congruence shortcut is known as side side side sss another shortcut is side angle side sas where two pairs of sides and the angle between them are known to be congruent

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web asa if two angles and the included side of one triangle are congruent to the corresponding parts of another triangle then the triangles are congruent sas if any two angles and the included side are the same in both

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web quiz 1 using either sss or sas determine which triangles are congruent quiz 2 to write the congruence statement match

the corresponding vertices since the side opposite r corresponding to the side opposite i r corresponding to i quiz 3 use what you have learned to solve for what is presented what are the sss and sas theorems

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web sal introduces and justifies the sss sas asa and aas postulates for congruent triangles he also shows that aaa is only good for similarity for ssa better to watch next video

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web nov 22 2022 ab qr bc rp and ca pq hence $\triangle abc \cong \triangle qrp$ sas side angle side in the postulate of sas in comparison among two triangles if two sides and the angle included between two sides are equal then it satisfies the condition of sas in the above given figure we can see that