

Thermodynamics (Chapter 1)

1.Introduction

Thermodynamics is the science of the relationship between heat, work and properties of the system. it is concerned with the means necessary to convert heat energy from the available sources such as chemical fuels and nuclear piles into mechanical work.

1.1 Defining terms:

System: may be defined as a collection of matter within prescribed and identifiable boundaries

Heat: is form of energy which is transferred from one body to another body that is at lower temperature, by virtue of temperature difference (gradient).

Work: is defined as the product of a force and the distance moved in the direction of the force. Work is observed to be energy in transition. It never contained in a body or passed by a body

NB Both heat and work are transitory energies and must be confused with the internal energy of the system

2. Thermodynamics Properties

These properties are any characteristic of a substance that can be observed. Examples being: pressure, volume, temperature, internal energy and enthalpy.

The thermodynamics properties which are independent of mass such as: temperature and pressure are categorized as intensive properties. While the properties that are dependent on mass, such as: volume and energy in its forms are categorized as extensive properties.

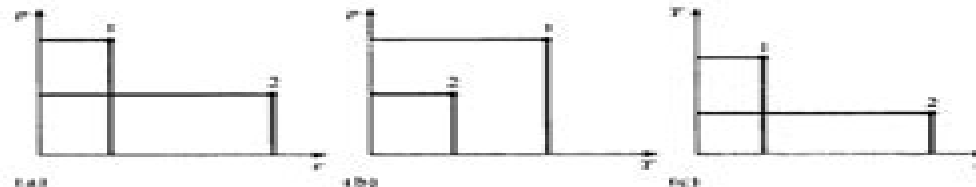
Intensive Properties

Temperature: is the measure of hotness and coldness of a system. It can also be defined as the average kinetic energy of molecules. The instrument used to measure temperature is commonly known as the thermometer.

3.The state of the working fluid

At any instant the state of the working fluid may be defined by certain characteristics called its properties. The thermodynamics properties introduced in this book are pressure, temperature, specific volume, specific internal energy, specific enthalpy and specific entropy. It is found that, for any pure working fluid, only two independent properties are necessary to define completely the state of a system, it is possible to represent the state of a system by a point situated on a diagram of properties. The illustration of diagram of properties is shown below.

Fig. 1.11 State of a working fluid on a property diagram



Study Guide For Thermodynamics Answers

Mr. Rohit Manglik



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Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

Chemical Principles Study Guide/Solutions Manual John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications

AP Biology Study Guide Sundar Nathan, 2009-11 Sundar Nathan received a Bachelor's degree in Electrical Engineering from Anna University Chennai India and a Masters degree in Biomedical Engineering from the University of Texas at Austin Working for over a year with a team of talented PhDs MPhils and MScs from all over the world Sundar compiled this comprehensive study guide to help students prepare diligently understand the concepts and Crush the AP Bio Test

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