

Matter Unit Study Guide

Atoms are everything that has mass, _____, and
 takes up space. _____
 Electromagnetic radiation: _____, _____, _____,

 Matter and energy are interchangeable due to E=mc²
 matter _____ energy and energy _____

States of matter of elements, molecules are listed
 the subatomic particles and their charge:



Subatomic particles and their
 charges:
 Electrons (negative)
 Protons (positive)
 Neutrons
 Quarks (fractional)
 Gluons (negative)

Classification of particles according to their
 interaction with the strong force: Quarks, leptons, and
 gauge bosons (including the Higgs boson).

1. Quarks: _____, _____, _____
 2. Leptons: _____, _____, _____
 3. Gauge bosons: _____, _____, _____

Properties of particles: mass, charge, spin,
 color, flavor, and other quantum numbers.

What is an example of an element?

What

What is an example of a compound?

Water

What is an example of a heterogeneous mixture?

Salad

What is an example of a homogeneous mixture?

Alloy

Study Guide Matter And Atom Answers

Marcel A. Müller



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