#### Study Guide for Chemistry EOC Assessment

Your test will consist of multiple-choice questions over the following topics.

- 1. Metric units for mass, length, volume
- Metric to metric conversions (K H Da m/l/g D C M)
- Know how to put numbers in scientific notation and take them out.
- 4. Significant Digit rules
- Difference between chemical and physical properties and changes. There will be multiple questions on this.
- 6. Ways to tell if a chemical change (chemical reaction) is taking place.
- 7. Pure substances and mixtures
  - a. heterogeneous mixture
  - b. homogeneous mixture
- 8. Density & how it relates to floating and sinking
- Use the equation for density
- Density = Mass/volume (g/ml) (g) (ml)
- Know the phases of matter and the characteristics of each phase.
- 11. Law of Conservation of Matter
- 12. What is a mole in chemistry?
- 13. Convert grams to moles, and moles to grams
- Define an atom
- 15. Structure of an atom and where each subatomic particle is located.
  - a. Nucleus (contains protons and neutrons)
  - b. Protons (positive)
  - c. Neutrons (neutral)
  - d. Electrons (negative)
- 16. Isotopes
- 17. Elements
- 18. If given atomic # and mass, find the number of protons and neutrons in an atom.
- 19. Find the number of valence electrons, and tell what it says about reactivity.
- 20. Periodic Table
  - a. Know about the families
  - b. Group vs. period
- 21. Know about the properties of metals and nonmetals.
  - a. Ductile, malleable, conductors
- 22. Ionic & Covalent Compounds

How do they form? (Transfer or gain electrons) Characteristics of Each

- Be able to name ionic compounds
- 24. Predict oxidation numbers
- 25. Chemical Equations:

Know the parts of an equation

5 main types of reactions-know about them and be able to identify them (Synthesis, Decomposition, Single Displacement, Double Displacement, Combustion)

26. Acids & Bases-

Macro and Microscopic properties of each

What ions do they form, how do they affect litmus, what is their pH range?

Identify examples of each- Mono, di, and polyprotic acids

Tell if they are strong or weak

Acid-base indicator

# **Study Guide For Chemistry Eoc**

**Lewis Morris** 

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been revised with an even sharper focus on practical skills and procedures The first half of the book is devoted to safe laboratory practice as well as purification and analytical techniques particularly spectroscopic analysis The second half contains step by step experimental procedures each one illustrating a basic principle or important reaction type Tried and tested over almost three decades over 100 validated experiments are graded according to their complexity and all are chosen to highlight important chemical transformations and to teach key experimental skills New sections cover updated health and safety guidelines additional spectroscopic techniques electronic notebooks and record keeping and techniques such as semi automated chromatography and enabling technologies such as the use of microwave and flow chemistry New experiments include transition metal catalysed cross coupling organocatalysis asymmetric synthesis flow chemistry and microwave assisted synthesis Key aspects of this third edition include Detailed descriptions of the correct use of common apparatus used in the organic laboratory Outlines of practical skills that all chemistry students must learn Highlights of aspects of health and safety in the laboratory both in the first section and throughout the experimental procedures Four new sections reflecting advances in techniques and technologies from electronic databases and information retrieval to semi automated chromatography More than 100 validated experiments of graded complexity from introductory to research level A user friendly experiment directory An instructor manual and PowerPoint slides of the figures in the book available on a companion website A comprehensive guide to contemporary organic chemistry laboratory principles procedures protocols tools and techniques Experimental Organic Chemistry Third Edition is both an essential laboratory textbook for students of chemistry at all levels and a handy bench reference for experienced chemists Chemistry Exam Study Guide Houk, 2018-12-03 EVERYTHING YOU NEED TO HELP SCORE A PERFECT SCORE Equip yourself to ace your Chemistry Exam with this comprehensive study guide including full length practice tests thorough content reviews targeted strategies for every section Books In Print 2004-2005 Ed Bowker Staff, Staff Bowker, Ed, 2004

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