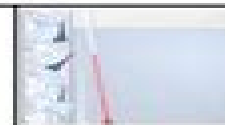


Activity A:
Phase changes

Get the Gizmo ready:

- Click **Reset** (↺) and select **Macro view**.
- Set the **Water temperature** to 10 °C.
- Set the **Ice volume** to 0 cm³.



Question: How is temperature related to phase changes?

1. **Predict:** Based on your prior knowledge, predict the following:

- A. At what temperature will water change from a liquid to a solid (**freeze**)?
- B. At what temperature will water change from a solid to a liquid (**melt**)?
- C. At what temperature will water change from a liquid to a gas (**boil**)?

0
0
100

2. **Investigate:** Use the Gizmo to explore phase changes. Use the **Add/remove heat energy** slider to control the water temperature. Record your observations in your notes, then answer the questions below:

- A. At what temperature does water freeze?
- B. At what temperature does ice melt?
- C. At what temperature does water boil?

0
0
10

- This is the **freezing point**.
- This is the **melting point**.
- This is the **boiling point**.

3. **Observe:** Set up the Gizmo to observe freezing. What do you notice about the temperature while the water is in the process of freezing?

Remains unchanged

4. **Explore:** Use the Gizmo to investigate melting and boiling. Does the temperature change while either of these phase changes is occurring?

No it doesn't

5. **Interpret:** Select the **GRAPH** tab to see a graph of temperature vs. time. Click the “–” button until the whole graph is visible. What does the graph look like during a phase change?

A horizontal line

6. **Extend your thinking:** Why do you think the temperature does not change much during a phase change? If possible, discuss your answer with your classmates and teacher.

Utilizing the energy to alter the nature of matter

Study Guide Phase Changes Answer Key

**Lisa Keenan-Lindsay, Cheryl A
Sams, Constance L. O'Connor, Shannon
E. Perry, Marilyn J. Hockenberry, Deitra
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