

**Section 2.1 – continued**

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

19
18
Hydrogen
1.008
Helium

Atomic Mass	1.01
Atomic Number	18
Electron Configuration	18
Chemical Name	18
Chemical Symbol	18

Put each item in Column A, under the letter of the matching item in Column B.

Column A

- _____ 19. It exists as the periodic table.
- _____ 20. It has no the periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are strong oxidizing ability.
- _____ 23. Group 18 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. noble gases

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 24. There are two representative elements of elements.
- _____ 25. Most of the elements of the periodic table are metals.
- _____ 26. Group 1 elements except for hydrogen are metals in the solid state.
- _____ 27. Group 18 elements are the noble gases.
- _____ 28. Group 1 elements are highly reactive elements in the periodic table.
- _____ 29. Group 18 elements are very reactive elements in the periodic table.
- _____ 30. Elements have properties of both metals and non-metals.

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RS Peters



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