



States of Matter: Phase Changes

Phase	Description	Illustration
Evaporation	Liquid → _____ ❖ Water changes into a gas called _____ ❖ Heat energy is added ❖ Particles move faster and faster ❖ Examples include a _____ disappearing and clothes drying	
Condensation	Gas → _____ ❖ Water vapor turns into droplets of (liquid) _____ ❖ Heat energy is removed ❖ Particles _____ ❖ Examples include morning dew and a foggy mirror after a hot shower	
Freezing	Liquid → _____ ❖ Heat energy is _____ ❖ (Liquid) water turns into ice ❖ Particles slow down ❖ Examples include rain turning into _____ and juice freezing into a popsicle	
Melting	Solid → _____ ❖ Heat energy is _____ ❖ Ice turns into (liquid) water ❖ Particles move _____ ❖ Examples include a snowman thawing out and wax dripping down a candle	

❖ Matter can change from one phase to another with the addition or removal of _____

❖ Matter changing from one state to another is a _____

Study Guide Phase Changes

**John Krenos, Joseph Potenza, Loretta
Jones, Lynn Koplitz, Thomas Spence**



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Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement **The**

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