CHEMISTRY PAPER 1 (THEORY)

Maximum Marks: 70

Time Allowed: Three hours

Candidates are allowed additional 15 minutes for only reading the paper.

They must NOT start writing during this time.

This paper is divided into four sections - A. B. C and D.

Answer all questions.

Section A consists of one question having sub-parts of one mark each.

Section B consists of ten questions of two marks each.

Section C consists of seven questions of three marks each, and

Section D consists of three questions of five marks each.

Internal choices have been provided in one question each in Section B, Section C and Section D.

All working, including rough work, should be done on the same sheet as, and adjacent to the rest of the answer.

The intended marks for questions or parts of questions are given in brackets []. Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

Gas constant R = 1.987 cal deg' mot' = 8.314 JK' mot' = 0.0821 dm' atm K'mot'

I I atm = I dm3 atm = 1013 J. I Faraday = 96500 coulombs.

Avogadro's number = 6.023×10^{19} .

SECTION A – 14 MARKS

Ouestion 1

(A) Fill in the blanks by choosing the appropriate word(s) from those given in the [4×1] brackets:

[stable, low, aldehyde, unstable, 6, 4, ethane, Clemmensen's, 2, 3, carboxylic acid, high, propane, Rosenmund's]

- (i) The primary alcohols are easily oxidised first into _____ and then into
- (ii) The intermediate activated complex in a chemical reaction is highly _____ due to _____ energy.
- (iii) The coordination number and oxidation state of the complex K₄[Fe(CN)₆] are _____ and ____ respectively.
- (iv) Propanone on reaction with zinc-amalgam in presence of conc. HCl gives and the reaction is known as _____ reduction.

This Paper consists of 8 printed pages.

Turn over

Vbcat Chemistry Question Paper

Arshad Iqbal

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organic chemistry into small segments Next 3 paper intermix two to three papers of the first eight paper and the last paper covers whole organic chemistry By doing this we want to revise your organic chemistry in 3 tiers so that not a single doubt should left out Organic chemistry is very easy and conceptual subject and need proper understanding of the basics and strategy to solve the questions in correct manner This Test paper will prepare your right mindset for learning Organic Chemistry This mindset is essentially the one that focuses you on a small number of straight forward fundamental concepts and helps you to apply them in different ways to solve the variety of problems you face in organic chemistry In this book balance has to be achieved between the number of questions and the quality of the questions especially because it is relatively easy to frame a very large number of multiple choice questions and theory of the subject The questions in this book have been selected keeping three things in mind First the questions are such that they really test the understanding of the subject Second the questions cover all concepts Third the number of questions has been kept large enough to offer meaningful practice to the students **Higher Certificate Chemistry Test Papers** Joseph Morris, 1935 Chemistry **Exam Elite** Chi Ming Lai, Yiu Chung Wong, 2021 Whitcombe's Chemistry Test Papers T. A. Glendinning, 1900 Junior Test Papers in Chemistry Eric John Holmyard, 1928 Higher certificate chemistry test papers J. Morris, 1930 Higher Certificate Chemistry Test Papers Joseph MORRIS (M.A. (Oxon.), of the Collegiate School, Liverpool.),1928 **Higher Level** Thinking Questions Angela Manzi, Miguel (CRT) Kagan, Michael Michels, 2009-12-17 Your secondary students will explore chemistry topics and themes like never before with this giant collection of ready to use chemistry questions You ll find questions for eighteen chemistry topics and themes Acids and Bases Atomic Structure Biochemistry Bonding Chemical Industry Chemical Reactions Electron Configurations Environmental Chemistry and Pollution Gases Hydrocarbons and Petroleum Metals Methods and Tools Nuclear Chemistry and Alternative Energy Sources Periodic Table Solutions States of Matter Stoichiometry and Water Nurture the development of your students thinking skills with hundreds of guestions such as How can gas pressure affect the health of someone who goes scuba diving Why does maple sugar flow from a tree in winter Promote thinking and active engagement in your chemistry class with this great book

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