

# 8 Structure and nomenclature of organic compounds

Unit	4	How are organic compounds categorised, analysed and used?
Area of Study	1	How can the diversity of carbon compounds be explained and categorised?
Topic	8	Structure and nomenclature of organic compounds
Subtopic	8.2	The carbon atom

## Summary

- Carbon atoms have 4 valence electrons and make 4 covalent bonds with other carbon atoms or atoms of different elements.
- Bond energy is the amount of energy required to break the bonds of reactants or the amount of energy released per mole when a covalent bond forms between 2 molecules. It is typically measure in  $\text{kJ mol}^{-1}$ .
- The bond length is the distance between the nuclei of the atoms sharing the electrons.
- Carbon molecules with single bonds typically form a tetrahedral arrangement.
- Double  $\text{C}=\text{C}$  bonds are stronger than  $\text{C}-\text{C}$  single bonds and forms flat molecules.  $\text{C}\equiv\text{C}$  bonds are stronger again and form linear molecules.



## My notes

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# Vcaa Chemistry Study Guide

**Richard Morris**



## Vcaa Chemistry Study Guide:

*Fortify VCE Chemistry Units 3 and 4 2017-2021 Study Guide* Joseph Rowan, Sophie Zhou, Emma Zheng, 2020-07

Jacaranda Chemistry 1 VCE Units 1 and 2, learnON and Print Neale Taylor, Angela Stubbs, Robert Stokes, 2022-11-30

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VCE Physical Education Mark Corrie, Sharon Teesdale, 2004

**VCE Chemistry; Study Guide to Unit 1**, 1995

Multimodal Literacy in School Science Len Unsworth, Russell Tytler, Lisl Fenwick, Sally Humphrey, Paul Chandler, Michele Herrington, Lam Pham, 2022-02-25

This book establishes a new theoretical and practical framework for multimodal disciplinary literacy MDL fused with the subject specific science pedagogies of senior high school biology chemistry and physics It builds a compatible alignment of multiple representation and representation construction approaches to science pedagogy with the social semiotic systemic functional linguistic based approaches to explicit teaching of disciplinary literacy

The early part of the book explicates the transdisciplinary negotiated theoretical underpinning of the MDL framework followed by the research informed repertoire of learning experiences that are then articulated into a comprehensive framework of options for the planning of classroom work

Practical adoption and adaptation of the framework in biology chemistry and physics classrooms are detailed in separate chapters The latter chapters indicate the impact of the collaborative research on teachers professional learning and students multimodal disciplinary literacy engagement

concluding with proposals for accommodating emerging developments in MDL in an ever changing digital communication world The MDL framework is designed to enable teachers to develop all students disciplinary literacy competencies

This book will be of interest to researchers teacher educators and postgraduate students in the field of science education It will also have appeal to those in literacy education and social semiotics

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*VCE Chemistry Study Guide to Unit 2*, 1992

A+ Chem Notes VCE Unit 3 Jennifer Willis, 2012

The third edition of Chem Notes VCE Unit 3 has been written to precisely match the VCE Chemistry Study Design to be implemented in 2013

This edition includes a data book and is 100% up to date Chem Notes VCE Unit 3 includes a Comprehensive summary notes on chemical analysis and organic chemical pathways a Practice exam questions a Solutions a Data sheets a Examination

advice a Glossary of terms a Revision checklists Chem Notes 3 has been designed to be the most comprehensive and easy to use study guide for students of VCE Chemistry Unit 3

*Valuing Assessment in Science Education: Pedagogy, Curriculum, Policy* Deborah Corrigan, Richard Gunstone, Alister Jones, 2013-06-05 Assessment is a fundamental issue in research in science education in curriculum development and implementation in science education as well as in science teaching and learning This book takes a broad and deep view of research involving assessment in science education across contexts and cultures from whole countries to individual classrooms and across forms and purposes from assessment in the service of student learning to policy implications of system wide assessment It examines the relationships between assessment measurement and evaluation explores assessment philosophies and practices in relation to curriculum and scientific literacy learning and details the relationships between assessment and science education policy The third in a series Valuing Assessment in Science Education has chapters from a range of international scholars from across the globe and staff from Monash University King s College London and University of Waikato The two previous books in the series examined research relevant to the re emergence of values in science education and teaching across the spectrum of science education as well as across cultural contexts through the professional knowledge of science teaching This third book now moves to examine different aspects of generating understanding about what science is learnt how it is learnt and how it is valued Valuing Assessment in Science Education will appeal to all those with some engagement with and or use of research in science education including research students academics curriculum development agencies assessment authorities and policy makers It will also be of interest to all classroom science teachers who seek to keep abreast of the latest research and development and thinking in their area of professional concern

**Jacaranda Maths Quest 12 Specialist Mathematics VCE Units 3 and 4 2e learnON and Print** Raymond Rozen, 2022-11-30 [VCE Chemistry Study Guide to Unit 3. Chemistry and the Market Place](#), 1995

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