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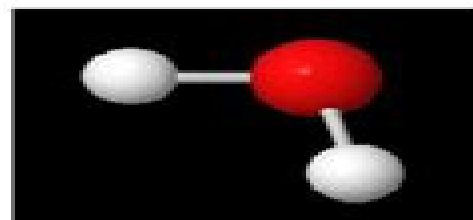
Student Exploration: Molecule Builder

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical bond, chemical formula, covalent bond, double covalent bond, cyclic molecule, hydrocarbon, ionic bond, isomers, Lewis structure, molecule, organic compound

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Look at the water **molecule**. In the molecule, **chemical bonds** are shown as lines between atoms.



1. The red atom in the middle is oxygen. How many bonds does oxygen have?

Two covalent bonds.

2. The two white atoms are hydrogens. How many bonds does one hydrogen atom have?

One

3. The **chemical formula** of water is H_2O . In this formula, what do you think the "2" between the H and O refers to?

Two hydrogen atoms

Gizmo Warm-up

In the Molecule Builder Gizmo, drag an oxygen atom into the white simulation area. The short blue lines extending from the atom represent possible chemical bonds.



1. How many bonds can oxygen form?

two

2. Drag a hydrogen atom into the area. How many bonds can oxygen form?

one

3. Drag the hydrogen atom on top of the oxygen atom. What happens?

They bond together.

The black line represents a chemical bond between the hydrogen and oxygen atoms.

4. Bond another hydrogen atom to the other side of the oxygen atom. Now, enter the formula "H₂O" into the **Type the formula below** box. Congratulations! You have created a water

Virtual Lab Exploration Guide

Scott C. Dulebohn

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Virtual Lab Exploration Guide

1. Understanding the eBook Virtual Lab Exploration Guide
 - The Rise of Digital Reading Virtual Lab Exploration Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Virtual Lab Exploration Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Virtual Lab Exploration Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Virtual Lab Exploration Guide
 - Personalized Recommendations
 - Virtual Lab Exploration Guide User Reviews and Ratings
 - Virtual Lab Exploration Guide and Bestseller Lists
5. Accessing Virtual Lab Exploration Guide Free and Paid eBooks

- Virtual Lab Exploration Guide Public Domain eBooks
- Virtual Lab Exploration Guide eBook Subscription Services
- Virtual Lab Exploration Guide Budget-Friendly Options
- 6. Navigating Virtual Lab Exploration Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Virtual Lab Exploration Guide Compatibility with Devices
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- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Virtual Lab Exploration Guide
 - Highlighting and Note-Taking Virtual Lab Exploration Guide
 - Interactive Elements Virtual Lab Exploration Guide
- 8. Staying Engaged with Virtual Lab Exploration Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Virtual Lab Exploration Guide
- 9. Balancing eBooks and Physical Books Virtual Lab Exploration Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Virtual Lab Exploration Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Virtual Lab Exploration Guide
 - Setting Reading Goals Virtual Lab Exploration Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Virtual Lab Exploration Guide
 - Fact-Checking eBook Content of Virtual Lab Exploration Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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