

Answers

Solids, Liquids, and Gases

Chapter Test A

1. B
2. C
3. A
4. D
5. C
6. B
7. C
8. C
9. B
10. C
11. gas
12. thermal
13. gas
14. viscosity
15. Boiling
16. F, more freely than
17. F, amorphous
18. F, freezing
19. F, sublimation
20. T
21. State B. The figure shows that the particles are arranged in a pattern of definite, fixed positions. They can vibrate in place but not move around.
22. State A. The particles are not in a fixed pattern as particles in a solid are; the particles can move around one another, but they remain quite close together.
23. State C, because heating increases the energy of the particles. As the temperature of the liquid increases, the particles would begin to vibrate more quickly. They would move faster and begin to move farther apart.
24. In a gas, particles are spread far apart and they move in all directions at high speeds, filling all the space available. In a liquid, particles are packed closely together but can move freely around one another. As a result, a liquid can take on the shape of its container. In a solid, particles are packed tightly together; they can vibrate in place but not move from their fixed positions.
25. In an ice cube, water molecules are arranged in a crystalline form. Inside the freezer, the ice cube is at a temperature below 0°C . When it is removed from the freezer, the ice cube absorbs thermal energy from its surroundings, mostly from the air in the room. At first, the added thermal energy causes the water molecules to vibrate faster, raising their temperature. When the temperature of the water molecules reaches 0°C , the melting point of water, the molecules vibrate so quickly that they break free from their fixed positions, and the ice begins to melt.
26. If the cylinder were heated, the temperature of the gas would increase. With the plunger held steady, the volume of the gas would remain constant. Under these conditions (increasing temperature and constant volume), the pressure of the gas would increase.

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Pearson

**Monica G. Turner, Robert H.
Gardner, Robert V. O'Neill**



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fingerprints bloodstains footwear impressions and tire track impressions The biological crime scene and recognizing collecting and preserving biological evidence including forensic entomology and evidence found at bioweapon scenes The fundamental principles of evidence as expressed by the Principle of Divisible Matter and the Locard Exchange Principle every touch leaves a trace Trace evidence including glass paint and soil Shooting incident scenes with discussion of bullet paths and gunshot residue The final section examines fire scenes quality assurance issues and methods for collecting and preserving various evidence types not covered in other chapters The delicate balance among logic science and investigative activity must be understood in order to successfully work a crime scene Enhanced by more than 200 color images this volume provides investigators and students with the tools to grasp these critical concepts paving an expeditious path to the truth

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on enabling function and enhancing participation rather than concentrating on dysfunction and disability A companion Evolve website includes all references linked to MEDLINE as well as helpful links to other relevant websites

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