# Valence Electrons

How many electrons are present in the following elements? [1]

Carbon \_\_\_ B Chlorine \_\_\_ a Indium \_\_\_ a Calcium \_\_\_

Using the periodic table, determine the number of valance electrons for the 2 elements listed below.

Barium In Iodine Carbon In Phosphorous

Circle the valence electrons only. What is the atomic number?









Atomic number(Z) = Atomic number(Z) = Atomic number(Z) = Atomic number(Z) =

Fill in the table with correct values. 4

| Element   | Ion symbol | Number of<br>Protons | Number of<br>electrons | Charge | Ion type |
|-----------|------------|----------------------|------------------------|--------|----------|
| Fluorine  | F-         |                      |                        | - 1    | Anion    |
|           |            | 53                   | 54                     |        |          |
|           |            | 16                   |                        | -2     | 1        |
| Potassium |            |                      |                        | +1     |          |
| Calcium   |            |                      |                        | .,     |          |

# **Valence Electrons Worksheet Answers Chemistry If8766**

G. Klopman, B. O'Leary

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