Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON METALS share electrons, attempting to attain a stable octet (8 outer electrons)in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

H_{2 brothrogen} is distornic

H + H → H → H-H Single Bond

2.) F2 thurston in discounts

3.) O2 oxygen is diatomic

O + O

 \bullet $\bigcirc =\bigcirc$

Double Bond

4.) No nitrogen is diatomic. Is this a triple bond?

N + N

5.) BF_{3 you need 3 fluorine atoms bere-}

 $\mathbf{R} + \mathbf{F}$

6.) Ammonia NH3 hist: how many hydrogen atoms are needed?

N + H

7.) Carbon dioxide CO₂

0 + c + 0

Methane CH₄ careful here 4 hydrogen atoms needed

C + H

9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!

H + O + H

10.) SO₂ hint: one pair of electrons from sulfur must be slit up for this one to work.

O + S + O

The Nature Of Covalent Bonding 8 2 Worksheet Answers

David Baud

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web fructose and lactose intolerances are common in fgid and associated with increased non gi symptoms but not with specific fgid subtypes symptoms experienced during breath testing but not malabsorption correlate with fgid symptoms effective symptom relief with dietary adaptation is not associated with malabsorption

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